

# Density Worksheet

## Physical Science

### $D = m/V$

Densities of Common Substances @ 20°C				
Substance	Density (g/cm <sup>3</sup> )		Substance	Density (g/cm <sup>3</sup> )
Oxygen	0.00143		Aluminum	2.70
Hydrogen	0.00008984		Iron	7.87
Ethanol	0.789		Copper	8.96
Benzene	0.884		Silver	10.5
Water	1.000		Lead	11.34
Magnesium	1.74		Mercury	13.6
Salt (sodium chloride)	2.16		Gold	19.32

- The ratio of an object's mass to its \_\_\_\_\_ is called the density of the object.
- A kilogram of lead occupies a much smaller volume than a kilogram of water, because \_\_\_\_\_ has a much higher density.
- For the masses and volumes indicated, calculate the density in grams per cubic centimeters.
  - mass = 453 g; volume = 225 cm<sup>3</sup>
  - mass = 5.0 g; volume = 10.0 cm<sup>3</sup>
  - mass = 26.1 g; volume = 2.0 mL
- If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.
- A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

# Half Life Calculations Physical Science Worksheet

## Answer Key

**Padhraic Smyth**



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