



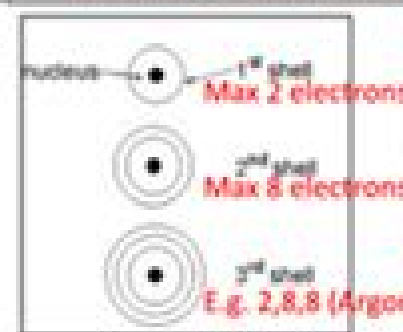
Periodic Table of the Elements

An atom:
Number of Protons = Number of Electrons
Mass number = Protons + Neutrons

C2 Chemistry

Periodic table arranges elements by **ATOMIC NUMBER** (proton number/the small one)

The number of outer shell electrons match the group the element is found in.
E.g. Lithium 2,1 is a group 1 element.

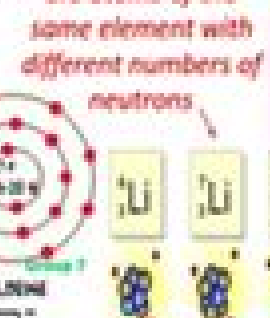
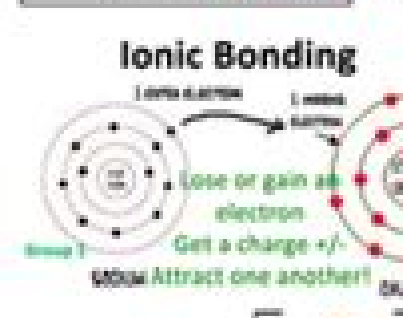
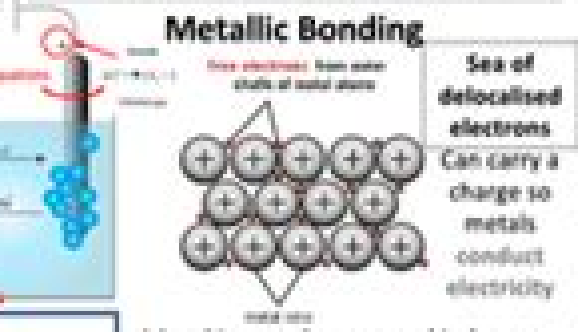
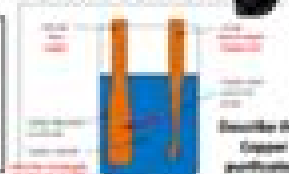


Protons and neutrons both have a mass of 1

ISOTOPES
are atoms of the same element with different numbers of neutrons

OIL RIG

Oxidation is loss (of electrons)
Reduction is gain (of electrons)



As a solid: High melting and Boiling Point
Conducts electricity when melted or dissolved: ions move freely and carry a charge

Metal ions slide over one another making them MALLEABLE

Metals conduct HEAT well

NANO-TECHNOLOGY
Means REALLY REALLY REALLY small!!!!

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Wenbin Ji



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