

PAPER 2 – STRUCTURE

NO.	Answer	Marks										
1(a)	Haber Process	1										
(b)	$N_2 + 3H_2 \rightarrow 2NH_3$	1										
(c)	Iron	1										
(d)	Temperature, 450°C ; Pressure 200 atm	1										
(e)	Ammonium sulphate	1										
1(b)(i)	$2NH_3 + CO_2 \rightarrow CO(NH_2)_2 + H_2O$	1										
(ii)	RMM urea = 60	1										
	% N = $\frac{28.02}{60} \times 100$	1										
	= 46.7%											
1(c)	1. Blue precipitate soluble in excess ammonia solution 2. Forming dark blue solution	2										
2(a)(i)	Contact process	1										
(ii)	Haber Process	1										
(b)(i)	$2SO_2 + O_2 \rightarrow 2SO_3$	1										
(ii)	Vanadium(V) oxide	1										
(c)(i)	$N_2 + 3H_2 \rightarrow 2NH_3$	1										
(ii)	Temperature, 450°C ; Pressure 200 atm	1										
2(b)(i)	Ammonium sulphate	1										
(ii)	$(NH_4)_2SO_4$											
(iii)	RMM ammonium sulphate = 132											
	% N = $\frac{28.02}{132} \times 100 = 21.2\%$	2										
(c)	Manufacture of detergents/ production of fertiliser/ manufacture of artificial fibres/ manufacture of paper / leather tanning/ as electrolyte in car batteries	1										
3(a)	Large molecules made up of many identical repeating sub-units called monomers which are joined together by covalent bonds.	1										
(b)	<table><tr><td>Monomer</td><td>Polymer</td></tr><tr><td>Amino acid</td><td>Protein</td></tr><tr><td>Glucose</td><td>Starch</td></tr><tr><td>Ethene</td><td>Polyethene</td></tr><tr><td>Isoprene</td><td>Natural rubber</td></tr></table>	Monomer	Polymer	Amino acid	Protein	Glucose	Starch	Ethene	Polyethene	Isoprene	Natural rubber	4
Monomer	Polymer											
Amino acid	Protein											
Glucose	Starch											
Ethene	Polyethene											
Isoprene	Natural rubber											
(c)	$\left[\begin{array}{cc} H & H \\ & \\ -C & -C- \\ & \\ H & H \end{array} \right]_n$											
3(b)(i)	PVC does not melt, iron will melt											
(ii)	$H - \overset{\overset{H}{ }}{C} = \overset{\overset{H}{ }}{C} - Cl$											
(c)	Natural polymer – protein, starch, natural rubber Synthetic polymer – polystyrene, polychloroethene											

Jan 2014 Wch03 01 Mark Scheme Chemistry

Iseb Iseb



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13+ Science Level 2 Chemistry Mark Scheme (Summer Term 2014) Iseb Iseb, 2014-03-28

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1. This book is structured into several chapters, namely:

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- Chapter 2: Essential Elements of Jan 2014 Wch03 01 Mark Scheme Chemistry
- Chapter 3: Jan 2014 Wch03 01 Mark Scheme Chemistry in Everyday Life
- Chapter 4: Jan 2014 Wch03 01 Mark Scheme Chemistry in Specific Contexts
- Chapter 5: Conclusion

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6. In chapter 5, this book will draw a conclusion about Jan 2014 Wch03 01 Mark Scheme Chemistry. This chapter will summarize the key points that have been discussed throughout the book.
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