Ideal Gas Law

The ideal gas law is the equation of state of an ideal gas that relates pressure, volume, quantity of gas, and absolute temperature.

PV = nRT

P = pressure
V = volume
n = number of moles of gas
R = ideal gas constant
T = absolute temperature

Although it applies to ideal gases, it approximates the behavior of many real gases.

Ideal Gas Laws Introduction

Arihant Experts

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