

Ideal Gas Law

Chapter 9

1. A gas at a pressure of 1 atm and a volume of 10 liters has a temperature of 25°C. What is the number of moles of the gas present?
2. A sample of gas has a volume of 5 liters, a pressure of 2 atm, and a temperature of 300 K. What is the number of moles of the gas present?
3. A gas has a pressure of 1.5 atm, a volume of 20 liters, and a temperature of 25°C. What is the number of moles of the gas present?
4. A gas at a volume of 30 liters and a temperature of 27°C has a pressure of 2 atm. What is the number of moles of the gas present?
5. A gas at a pressure of 1 atm, a volume of 25 liters, and a temperature of 0°C has a number of moles equal to 2. What is the pressure of the gas at a volume of 15 liters and a temperature of 27°C?
6. A gas at a pressure of 2 atm, a volume of 10 liters, and a temperature of 25°C has a number of moles equal to 4. What is the pressure of the gas at a volume of 20 liters and a temperature of 27°C?
7. A gas at a volume of 5 liters and a temperature of 27°C has a pressure of 1 atm. What is the number of moles of the gas present?
8. A gas at a pressure of 3 atm, a volume of 20 liters, and a temperature of 25°C has a number of moles equal to 5. What is the pressure of the gas at a volume of 10 liters and a temperature of 27°C?
9. A sample of gas has a volume of 15 liters, a pressure of 1.5 atm, and a temperature of 300 K. What is the number of moles of the gas present?
10. A gas at a volume of 10 liters and a temperature of 27°C has a pressure of 2 atm. What is the number of moles of the gas present?
11. Note: These questions are meant to be examples to guide you in creating your own worksheet. The specific answers to these questions may vary based on your textbook or course material.

Ideal Gas Law Questions And Answers

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