

ICP: Chemistry Review

0

out of 150

Directions: In order to earn you credit for 1st semester ICP, you must complete the following assignments. When you have completed 100% of the work below, the assignment is considered finished. When answering questions, if the answer is correct, the cell will turn color! For your reference, a periodic table is located [on the next tab](#).

Lab Equipment and Lab Safety

Directions for #1-5, identify the following pieces of lab equipment. Once all are correct, the cells will turn green. Then, decide if the statements below are true or false by choosing yes or no. Once all the answers in a column are correct, the cells will turn green.



1



3



4



5



Yes or No? (You must answer all questions in the column correctly)

Yes or No? (You must answer all questions in the column correctly)

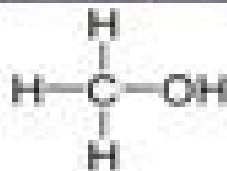
- 6. Wear safety goggles when using chemicals, hot liquids, or flames
- 7. You can try new experiments even if they are not in the directions as long as you are curious about science
- 8. Tie back long hair when using chemicals or open flames
- 9. If you spill any material, quickly clean it up with a cloth before the teacher can see that you spilled it.
- 10. When you have finished an experiment, leave your work area clean and dump all of your chemicals down the sink

- 11. It is ok to leave a burner flame alone for a few seconds while you grab some materials
- 12. In an experiment, you should have on proper lab clothing including close-toed shoes.
- 13. You should always wash your hands before an experiment, but you can decide if you need to wash your hands after an experiment
- 14. To safely smell a chemical in lab you should inhale the fumes directly from the container
- 15. Food and gum are NOT allowed in the lab area at any time

16. Protons, Neutrons, and Electrons practice- fill in the missing information in the table below. When a row is correct, it will change color.

Atomic Symbol	Atomic Number	Protons	Neutrons	Electrons	Atomic Mass
C			6		
	11				23
		30	35		
	9		10		

Look at the diagram of methanol below to help you answer questions #17-19



- 17. How many atoms are in a compound of methanol?
- 18. What is the chemical formula of methanol?
- 19. What is the molar mass of methanol?

For #20-23, identify each as either a Physical or Chemical Change.

- 20. A phase change (ice to melting)
- 21. Mixing 2 solutions produces a substance with a new color or turns cloudy
- 22. A change that is easier to reverse
- 23. Dissolving sugar into water

Icp Final Review Chemistry Id A

David Z. Piper, Caroline M. Isaacs



Icp Final Review Chemistry Id A:

This book delves into Icp Final Review Chemistry Id A. Icp Final Review Chemistry Id A is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Icp Final Review Chemistry Id A, encompassing both the fundamentals and more intricate discussions.

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- Chapter 1: Introduction to Icp Final Review Chemistry Id A
- Chapter 2: Essential Elements of Icp Final Review Chemistry Id A
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- Chapter 4: Icp Final Review Chemistry Id A in Specific Contexts
- Chapter 5: Conclusion

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