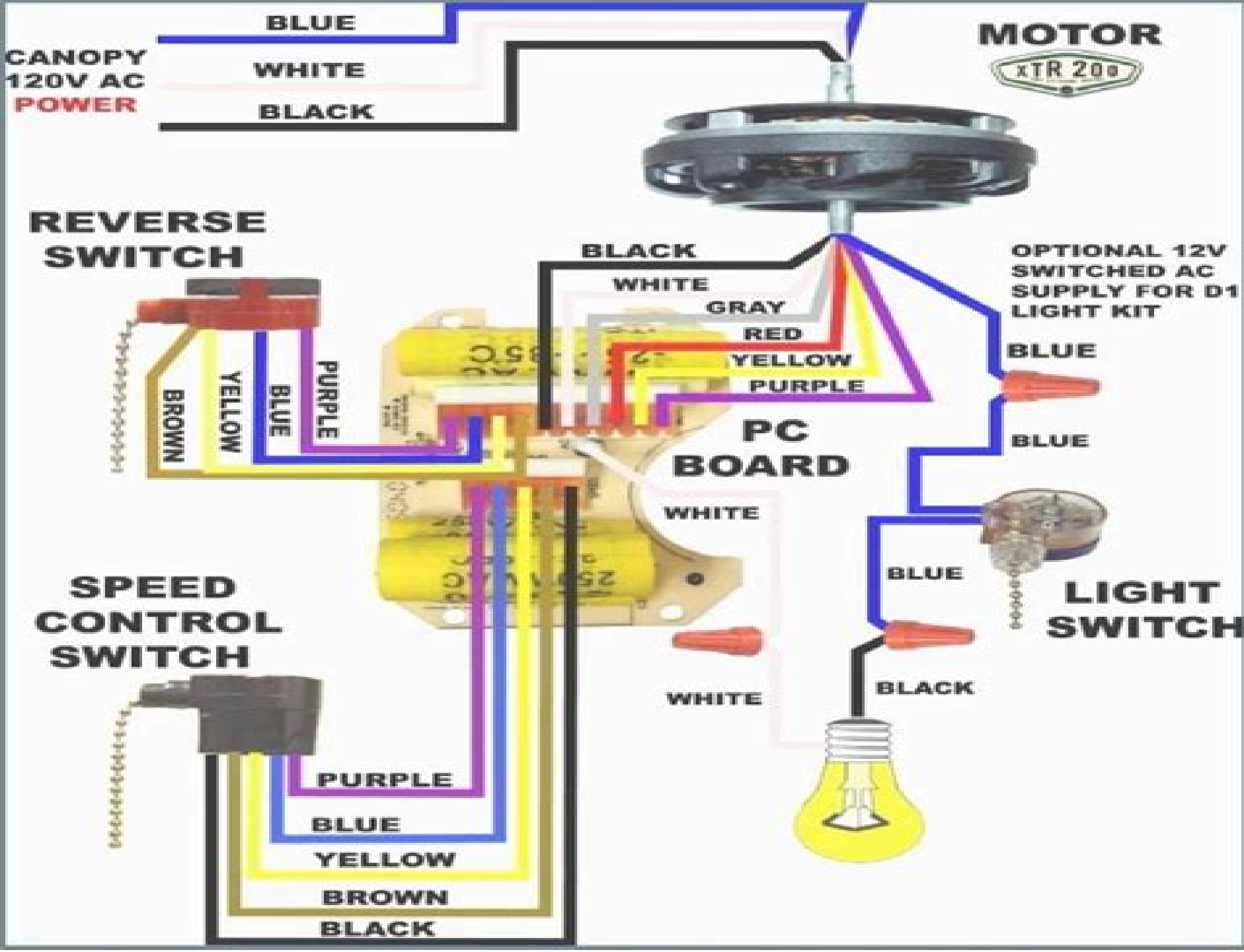




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web donor whereas a base generically represented by b is a proton acceptor when acids and bases combine they swap a proton $h_a + b \rightleftharpoons h_b + a^-$ strong acids e.g. sulfuric acid have a very pronounced tendency to transfer a proton to a base whereas weak acids

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web 1 approximately 3 g of hydrated oxalic acid $H_2C_2O_4 \cdot 2H_2O$ was weighted to the nearest 0.01 g in a 50 mL beaker 2 about

30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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