

SECTION - A

Q. 1. Select and write the correct answer for the following multiple choice type of questions :

[10]

(i) The product obtained in the following reaction

P 239
11.2



- (a) Pent-3-en-1-ol ✓(b) Pentan-1-ol
(c) Pentan-2-ol (d) Pentanal

(ii) Amongst the following, the solubility of which ionic solid decreases with increase in temperature?

P 30
11)

- (a) KNO_3 (b) NaBr
✓(c) Na_2SO_4 (d) KCl

(iii) The correct IUPAC name of $\text{Na}_3 [\text{AlF}_6]$ is

P 196
Table
9.3

- ✓(a) Sodium hexafluoroaluminate (III)
(b) Sodium hexafluoroaluminate (II)
(c) Sodium hexafluoroaluminium (III)
(d) Sodium hexafluoroaluminium (II)

(iv) Which of the following acids has highest pKa value?

P 277
Table
12.9

- (a) Mono chloroacetic acid (b) Dichloroacetic acid
(c) Trichloroacetic acid ✓(d) Acetic acid

(v) Number of carbon atoms present in isoprene unit is ____.

P 322
15.3

- (a) 6 ✓(b) 5
(c) 4 (d) 3

(vi) The colourless transition metal ion amongst the following is

P 173

- (a) Cu^+ (b) Cu^{++}
(c) Ni^{++} (d) Co^{++}

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Scott C. Dulebohn



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