

1 Temperature**BEFORE YOU READ**

After you read this section, you should be able to answer these questions:

- How are temperature and kinetic energy related?
- How is temperature measured?
- What is thermal expansion?

National Science
Education Standards
PS 1a, 1b

What Is Temperature?

You may think of temperature as how hot or cold something is. But using the words *hot* and *cold* can be confusing. Pretend that you are outside on a hot day. If you step onto a porch with a fan blowing, you might think it feels cool. Then, your friend enters the porch from an air-conditioned house. She thinks the porch is warm!

Using the words *cool* and *warm* to describe the porch is confusing. Measuring the temperature on the porch tells you exactly how hot or cold it is. The **temperature** tells you the average kinetic energy of the particles in an object. **50**

TEMPERATURE AND KINETIC ENERGY

All matter is made of atoms or molecules that are always moving. The particles that are moving have kinetic energy. The faster they move, the more kinetic energy they have. Look at the figure below. The more kinetic energy the particles have, the higher the temperature is.



The gas particles on the right have higher kinetic energy than those on the left.

STUDY TIP

Describe Describe how temperature affects the average kinetic energy and thermal expansion. Give several examples of thermal expansion.

READING CHECK

1. **Identify** What is used to tell how hot or cold something is?

TAKE A LOOK

2. **Identify** Which gas shows particles at the higher temperature?

Holt Science Heat And Heat Technology

Rachel Sandford



Holt Science Heat And Heat Technology:

Holt Science Heat And Heat Technology Book Review: Unveiling the Power of Words

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Table of Contents Holt Science Heat And Heat Technology

1. Understanding the eBook Holt Science Heat And Heat Technology
 - The Rise of Digital Reading Holt Science Heat And Heat Technology
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Science Heat And Heat Technology
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Science Heat And Heat Technology
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Science Heat And Heat Technology
 - Personalized Recommendations
 - Holt Science Heat And Heat Technology User Reviews and Ratings

- Holt Science Heat And Heat Technology and Bestseller Lists
- 5. Accessing Holt Science Heat And Heat Technology Free and Paid eBooks
 - Holt Science Heat And Heat Technology Public Domain eBooks
 - Holt Science Heat And Heat Technology eBook Subscription Services
 - Holt Science Heat And Heat Technology Budget-Friendly Options
- 6. Navigating Holt Science Heat And Heat Technology eBook Formats
 - ePub, PDF, MOBI, and More
 - Holt Science Heat And Heat Technology Compatibility with Devices
 - Holt Science Heat And Heat Technology Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Science Heat And Heat Technology
 - Highlighting and Note-Taking Holt Science Heat And Heat Technology
 - Interactive Elements Holt Science Heat And Heat Technology
- 8. Staying Engaged with Holt Science Heat And Heat Technology
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Science Heat And Heat Technology
- 9. Balancing eBooks and Physical Books Holt Science Heat And Heat Technology
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Science Heat And Heat Technology
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Holt Science Heat And Heat Technology
 - Setting Reading Goals Holt Science Heat And Heat Technology
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Holt Science Heat And Heat Technology
 - Fact-Checking eBook Content of Holt Science Heat And Heat Technology
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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