

**NEB CLASS 11**  
**MODEL QUESTION**  
**(EFFECTIVE FROM 2077)**

Grade XI

Subject Code : Che. 301

Time : 3 hours

Attempt all questions

Subject: Chemistry (Theory)

Full marks: 75

**Group 'A'**

Circle the best alternative to the following questions.

(11×1 = 11)

1. How many atoms are there in two molecules of water?
  - a. 3
  - b. 4
  - c. 5
  - d. 6
2. What is the number of moles of ammonia gas formed when 0.5 mole of nitrogen gas is reacted with excess of hydrogen gas?
  - a. 0.5
  - b. 1
  - c. 2
  - d. 3
3. Which of the following bonding is responsible for the solubility of ammonia gas in water?
  - a. Hydrogen bonding
  - b. Ionic bonding
  - c. Covalent bonding
  - d. Van der Waals' force
4. What happens when Sulphur dioxide ( $\text{SO}_2$ ) gas is passed through an acidified solution of hydrogen sulfide ( $\text{H}_2\text{S}$ ) gas?
  - a.  $\text{SO}_2$  is oxidized to Sulphur
  - b.  $\text{H}_2\text{S}$  is reduced to Sulphur
  - c.  $\text{SO}_2$  is oxidized to  $\text{H}_2\text{SO}_4$
  - d.  $\text{SO}_2$  is reduced to Sulphur

# Grade Chemistry Question Paper June 23

**RS Peters**



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## **Table of Contents Grade Chemistry Question Paper June 23**

1. Understanding the eBook Grade Chemistry Question Paper June 23
  - The Rise of Digital Reading Grade Chemistry Question Paper June 23
  - Advantages of eBooks Over Traditional Books
2. Identifying Grade Chemistry Question Paper June 23
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Grade Chemistry Question Paper June 23
  - User-Friendly Interface
4. Exploring eBook Recommendations from Grade Chemistry Question Paper June 23
  - Personalized Recommendations
  - Grade Chemistry Question Paper June 23 User Reviews and Ratings
  - Grade Chemistry Question Paper June 23 and Bestseller Lists
5. Accessing Grade Chemistry Question Paper June 23 Free and Paid eBooks

- Grade Chemistry Question Paper June 23 Public Domain eBooks
- Grade Chemistry Question Paper June 23 eBook Subscription Services
- Grade Chemistry Question Paper June 23 Budget-Friendly Options
- 6. Navigating Grade Chemistry Question Paper June 23 eBook Formats
  - ePub, PDF, MOBI, and More
  - Grade Chemistry Question Paper June 23 Compatibility with Devices
  - Grade Chemistry Question Paper June 23 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Grade Chemistry Question Paper June 23
  - Highlighting and Note-Taking Grade Chemistry Question Paper June 23
  - Interactive Elements Grade Chemistry Question Paper June 23
- 8. Staying Engaged with Grade Chemistry Question Paper June 23
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Grade Chemistry Question Paper June 23
- 9. Balancing eBooks and Physical Books Grade Chemistry Question Paper June 23
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Grade Chemistry Question Paper June 23
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Grade Chemistry Question Paper June 23
  - Setting Reading Goals Grade Chemistry Question Paper June 23
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade Chemistry Question Paper June 23
  - Fact-Checking eBook Content of Grade Chemistry Question Paper June 23
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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