

## CHEMISTRY PAPER-II

## OBJECTIVE

TIME ALLOWED: 20 Minutes

MAXIMUM MARKS: 17

**Note:** You have four choices for each objective type question as A, B, C and D. The choice which you think is correct, fill that bubble in front of that question number, on bubble sheet. Use marker or pen to fill the bubbles. Cutting or filling two or more bubbles will result in zero mark in that question. No credit will be awarded in case BUBBLES are not filled. Do not solve question on this sheet of OBJECTIVE PAPER.

Q.No.1

- (1) Which one of the following does not belong to alkaline earth metals?  
 (A) *Be* (B) *Ra* (C) *Ba* ☒ (D) *Rn*
- (2) Which element forms an ion with charge +3?  
 (A) *Be* ☒ (B) *Al* (C) *C* (D) *Si*
- (3) Oxidation of *NO* in air produces:  
 (A) *N<sub>2</sub>O* (B) *N<sub>2</sub>O<sub>3</sub>* ☒ (C) *N<sub>2</sub>O<sub>4</sub>* (D) *N<sub>2</sub>O<sub>5</sub>*
- (4) Chlorine heptaoxide (*Cl<sub>2</sub>O<sub>7</sub>*) reacts with water to form:  
 (A) Hypochlorous acid (B) Chloric acid ☒ (C) Perchloric acid (D) Chlorine and oxygen
- (5) The anhydride of *HClO<sub>4</sub>* is:  
 (A) *ClO<sub>3</sub>* (B) *ClO<sub>2</sub>* (C) *Cl<sub>2</sub>O<sub>3</sub>* ☒ (D) *Cl<sub>2</sub>O<sub>7</sub>*
- (6) Which of the following is a non-typical transition element?  
 (A) *Cr* (B) *Mn* ☒ (C) *Zn* (D) *Fe*
- (7) Ethers show the phenomenon of:  
 (A) Position isomerism (B) Functional group isomerism ☒ (C) Metamerism (D) Cis-trans isomerism
- (8) Preparation of vegetable ghee involves:  
 (A) Halogenation ☒ (B) Hydrogenation (C) Hydroxylation (D) Dehydrogenation
- (9) Which of the following acid can be used as a catalyst in Friedal-Crafts reactions?  
☒ (A) *AlCl<sub>3</sub>* (B) *HNO<sub>3</sub>* (C) *BeCl<sub>2</sub>* (D) *NaCl*
- (10) For which mechanisms, the first step involved is the same?  
 (A) *E<sub>1</sub>* and *E<sub>2</sub>* (B) *E<sub>2</sub>* and *S<sub>N</sub>2* (C) *S<sub>N</sub>1* and *E<sub>2</sub>* ☒ (D) *E<sub>1</sub>* and *S<sub>N</sub>1*
- (11) Which compound is more soluble in water?  
☒ (A) *C<sub>2</sub>H<sub>5</sub>OH* (B) *C<sub>6</sub>H<sub>5</sub>OH* (C) *CH<sub>3</sub>COCH<sub>3</sub>* (D) *n*-Hexanol
- (12) Which of the following will have the highest boiling point?  
 (A) Methanal (B) Ethanal (C) Propanal ☒ (D) 2-Hexanone
- (13) Which of the following reagents will react with both aldehydes and ketones?  
☒ (A) Grignard's reagent (B) Tollen's reagent (C) Fehling's reagent (D) Benedict's reagent
- (14) Which reagent is used to reduce a carboxylic group to an alcohol?  
 (A)  $\text{H}_2/\text{Ni}$  (B)  $\text{H}_2/\text{Pt}$  (C) *NaBH<sub>4</sub>* ☒ (D) *LiAlH<sub>4</sub>*
- (15) Alkanenitriles can be prepared by treating alkyl halide with:  
 (A) Alcoholic *KOH* ☒ (B) Alcoholic *KCN* (C) Aqueous *KOH* (D) Aqueous *KNO<sub>3</sub>*
- (16) Most concentrated solid nitrogen fertilizer is:  
 (A) *NH<sub>3</sub>* (B) *(NH<sub>4</sub>)<sub>2</sub>HPO<sub>4</sub>* ☒ (C) *(NH<sub>2</sub>)<sub>2</sub>CO* (D) *NH<sub>4</sub>NO<sub>3</sub>*
- (17) Mark the correct statement.  
☒ (A) *Na<sup>+</sup>* is smaller than *Na* atom (B) *Na<sup>+</sup>* is larger than *Na* atom  
 (C) *Cl<sup>-</sup>* is smaller than *Cl* atom (D) *Cl<sup>-</sup>* ion and *Cl* atom are equal in size

# Mcqs Of Chemistry With Answers

**Arshad Iqbal**



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### **Table of Contents Mcqs Of Chemistry With Answers**

1. Understanding the eBook Mcqs Of Chemistry With Answers
  - The Rise of Digital Reading Mcqs Of Chemistry With Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Mcqs Of Chemistry With Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Mcqs Of Chemistry With Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Mcqs Of Chemistry With Answers
  - Personalized Recommendations
  - Mcqs Of Chemistry With Answers User Reviews and Ratings
  - Mcqs Of Chemistry With Answers and Bestseller Lists

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  - Mcqs Of Chemistry With Answers Enhanced eBook Features
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  - Highlighting and Note-Taking Mcqs Of Chemistry With Answers
  - Interactive Elements Mcqs Of Chemistry With Answers
8. Staying Engaged with Mcqs Of Chemistry With Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Mcqs Of Chemistry With Answers
9. Balancing eBooks and Physical Books Mcqs Of Chemistry With Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Mcqs Of Chemistry With Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Mcqs Of Chemistry With Answers
  - Setting Reading Goals Mcqs Of Chemistry With Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mcqs Of Chemistry With Answers
  - Fact-Checking eBook Content of Mcqs Of Chemistry With Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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