

21. In which of the following reactions addition does not take place according to Markovnikov's rule ?
(IIT 2000)
- (a) $\text{CH}_3\text{CH}=\text{CH}_2 + \text{HCl} \xrightarrow{\text{ROOR}}$
(b) $\text{CH}_3\text{CH}=\text{CH}_2 + \text{HBr} \xrightarrow{\text{ROOR}}$
(c) $\text{CH}_2=\text{CH}_2 + \text{HI} \xrightarrow{\text{ROOR}}$
(d) $\text{CH}_3\text{CH}=\text{CHCH}_3 + \text{HBr} \xrightarrow{\text{ROOR}}$
22. Ozonolysis of 2-methyl butene-2 yields : (UPC 2003)
- (a) ketone (b) aldehyde
(c) aldehyde & ketone (d) none
23. C_6H_{12} on ozonolysis gave only propanone. Alkene is :
(AFMC 2003)
- (a) $\text{CH}_3\text{CH}_2\text{CH}=\text{C}(\text{CH}_3)_2$
(b) $(\text{CH}_3)_2\text{C}=\text{CH}-\text{CH}_3$
(c) $(\text{CH}_3)_2\text{C}=\text{C}(\text{CH}_3)_2$
(d) $\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}_2\text{CH}_3$
24. The reaction,
 $\text{CH}_2=\text{CHCH}_3 + \text{HBr} \longrightarrow \text{CH}_3\text{CHBrCH}_3$ is :
(AIIMS 2003)
- (a) Nucleophilic addition
(b) Electrophilic addition
(c) Electrophilic substitution
(d) Free radical addition
25. Which of the following alkenes is most stable ?
(UPC 2003)
- (a) $\text{R}_2\text{C}=\text{CR}_2$ (b) $\text{R}-\text{CH}=\text{CH}-\text{R}$
(c) $\text{RCH}=\text{CH}_2$ (d) $\text{CH}_2=\text{CH}_2$
26. A molecule of HCl adds on 3-methyl but-1-ene. Product formed is : (IIT 2001)
- (a) $\text{CH}_3\text{CH}(\text{CH}_3)-\text{CHCl}-\text{CH}_3$
(b) $\text{CH}_3-\text{C}(\text{Cl})(\text{CH}_3)-\text{CH}_2-\text{CH}_3$
(c) $\text{CH}_3\text{CH}(\text{CH}_3)-\text{CH}_2-\text{CH}_2\text{Cl}$
(d) $\text{CH}_3\text{CH}(\text{CH}_3)-\text{CH}(\text{Cl})-\text{CH}_2\text{CH}_3$
27. The intermediate during the addition of HCl to propene in presence of peroxide is : (IIT 2002)
- (a) $\text{CH}_3\dot{\text{C}}\text{HCH}_2\text{Cl}$ (b) $\text{CH}_3\dot{\text{C}}\text{HCH}_3$
(c) $\text{CH}_3\text{CH}_2\dot{\text{C}}\text{H}_2$ (d) $\text{CH}_3\text{CH}_2\dot{\text{C}}\text{H}$
28. Which of the following exhibit *cis, trans*-isomerism ?
(a) $\text{CH}_2\text{Br}-\text{CH}_2\text{Br}$
- (b) $\text{CBr}_2=\text{CH}_2$
(c) $\text{CHBr}=\text{CHBr}$
(d) $\text{CBr}_2=\text{CH}-\text{CH}_3$
29. The compound that dehydrates most readily is :
(CPMT 2001)
- (a) R_2CHOH (b) RCH_2OH
(c) R_3COH (d) $\text{R}-\text{OH}$
30. The addition of propene with HOCl proceeds via the addition of : (IIT 2001)
- (a) H^+ in the first step
(b) Cl^+ in the first step
(c) OH^- in the first step
(d) Cl^+ and OH^- in a single step
31. Which compound is sure to be alkene ?
(AFMC 2003)
- (a) C_2H_4 (b) C_3H_6
(c) C_4H_8 (d) C_5H_{10}
(e) All
32. 1-chlorobutane with alc. KOH gives : (AFMC 2001)
- (a) 1-butene (b) 1-butanol
(c) 2-butene (d) 2-butanol
33. How much ethylene is oxidised by 3 litres of O_2 under identical conditions ? [CBSE (Med.) 2002]
- (a) One mole (b) One litre
(c) 3 litre (d) 0.3 litre
34. Which one of the following alkenes will react fastest with H_2 under catalytic hydrogenation condition ?
(IIT (S) 2000)
- (a) $\text{R}_2\text{C}=\text{CH}_2$ (b) $\text{RHC}=\text{CHR}$
(c) $\text{R}_2\text{C}=\text{CHR}$ (d) $\text{R}_2\text{C}=\text{CR}_2$
35. Which of the following gases gives a red precipitate with ammonical solution of cuprous chloride ?
(DCE 2004)
- (a) C_2H_4 (b) C_2H_6
(c) C_2H_2 (d) $\text{CH}_3\text{CH}_2\text{CH}_3$
36. Which will not give a precipitate with Tollen's reagent?
(DCE 2002)
- (a) Ethyne
(b) 2-pentyne
(c) 1-butyne
(d) 3-methyl-1-butyne
37. Which hydrocarbon reacts with sodium in liquid NH_3 ?
(AIEEE 2008)
- (a) $\text{CH}_3-\text{CH}_2-\text{C}\equiv\text{CH}$
(b) $\text{CH}_3-\text{C}\equiv\text{C}-\text{CH}_3$
(c) $\text{CH}_3-\text{CH}=\text{CH}-\text{CH}_3$
(d) $\text{CH}_3-\text{CH}_2-\text{CH}=\text{CH}_2$
38. Increasing order of volatility of C_2H_6 , C_2H_4 , C_2H_2 and C_6H_6 is :
(BHU 2003)

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of Green Chemistry the latest trend for sustainable chemistry as Appendix II

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This book delves into Mcqs Of Hydrocarbons. Mcqs Of Hydrocarbons is a crucial topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Mcqs Of Hydrocarbons, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Mcqs Of Hydrocarbons
- Chapter 2: Essential Elements of Mcqs Of Hydrocarbons
- Chapter 3: Mcqs Of Hydrocarbons in Everyday Life
- Chapter 4: Mcqs Of Hydrocarbons in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Mcqs Of Hydrocarbons. This chapter will explore what Mcqs Of Hydrocarbons is, why Mcqs Of Hydrocarbons is vital, and how to effectively learn about Mcqs Of Hydrocarbons.
3. In chapter 2, the author will delve into the foundational concepts of Mcqs Of Hydrocarbons. The second chapter will elucidate the essential principles that need to be understood to grasp Mcqs Of Hydrocarbons in its entirety.
4. In chapter 3, this book will examine the practical applications of Mcqs Of Hydrocarbons in daily life. The third chapter will showcase real-world examples of how Mcqs Of Hydrocarbons can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Mcqs Of Hydrocarbons in specific contexts. The fourth chapter will explore how Mcqs Of Hydrocarbons is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Mcqs Of Hydrocarbons. The final chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Mcqs Of Hydrocarbons.

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