May 2009 TZ1 Question 1

(b) For part of her extended essay investigation into the efficiency of the process, a student reacted a pure sample of a vegetable oil (where R=C₁,H₂₀) with methanol. The raw data recorded for the reaction is below.

| Mass of oil | 100 | 1013.0 g |
|--------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|------|----------|
| and and the control of the control o | - | 200.0 g |
| Mass of sodium hydroxide | - | 3.5 g |
| Mass of biodiesel produced | 1000 | 811.0 g |

(in moles) of the oil and the methanol used, and hence the amount (in moles) of excess methanol.

The relative molecular mass of the oil used by the student is 885.6. Calculate the amount

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May 2009 TZ2 Question 1

(b) A student reacted some salicylic acid with excess ethanoic anhydride. Impure solid aspirin was obtained by filtering the reaction mixture. Pure aspirin was obtained by recrystallization. The following table shows the data recorded by the student.

| Mass of salicylic acid used | $3.15 \pm 0.02 \text{ g}$ |
|-------------------------------|---------------------------|
| Mass of pure aspirin obtained | 2.50 ± 0.02 g |

- (i) Determine the amount, in mol, of salicylic acid, C_sH_s(OH)COOH, used. [2]
- (ii) Calculate the theoretical yield, in g, of aspirin, C_sH_s(OCOCH_s)COOH. /2/

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Chemistry for the IB Diploma Geoff Neuss, 2001 This concise guide provides the content needed for the Chemistry IB diploma at both Standard and Higher Level It follows the structure of the IB Programme exactly and includes all the options Each topic is presented on its own page for clarity Higher Level material is clearly indicated and there are plenty of practice questions The text is written with an awareness that English might not be the reader s first language Monthly Catalog of United States Government Publications .1962 Engineering World .1909 1 & EC ,1925 **Journal** of the Society of Chemical Industry Society of Chemical Industry (Great Britain),1892 The Publisher ,1911 **Papers Presented at the National Technical Conference** Society of Plastics Engineers, 1969 Bibliography of Agriculture Publications of the National Bureau of Standards United States. National Bureau of .1964 **Bulletin** .1910 Bulletin United States. Bureau of Mines, British Books ,1911 **Introduction to Green** Standards, 1983 **Chemistry** Albert Matlack, 2010-04-05 The book covers traditional green chemistry topics including catalysis benign solvents and alternative feedstocks It also discusses relevant but less frequently covered topics with chapters such as Chemistry of Longer Wear and Population and the Environment This coverage highlights the importance of chemistry to everyday life and demonstrates the benefits the expanded exploitation of green chemistry can have for society Copiously illustrated with over 800 figures this second edition provides an update from the frontiers of the field Introduction to Green Chemistry John Andraos, Albert S. Matlack, 2022-03-09 Interest in green chemistry and clean processes has grown so much in recent years that topics such as fluorous biphasic catalysis metal organic frameworks and process intensification which were barely mentioned in the First Edition have become major areas of research In addition government funding has ramped up the development of fuel cells and biofuels This reflects the evolving focus from pollution remediation to pollution prevention Copiously illustrated with more than 800 figures the Third Edition provides an update from the frontiers of the field It features supplementary exercises at the end of each chapter relevant to the chemical examples introduced in each chapter Particular attention is paid to a new concluding chapter on the use of green metrics as an objective tool to demonstrate proof of synthesis plan efficiency and to identify where further improvements can be made through fully worked examples relevant to the chemical industry NEW AND EXPANDED RESEARCH TOPICS Metal organic frameworks Metrics Solid acids for alkylation of isobutene by butanes Carbon molecular sieves Mixed micro and mesoporous solids Organocatalysis Process intensification and gas phase enzymatic reactions Hydrogen storage for fuel cells Reactive distillation Catalysts in action on an atomic scale UPDATED AND EXPANDED CURRENT EVENTS TOPICS Industry resistance to inherently safer chemistry Nuclear power Removal of mercury from vaccines Removal of mercury and lead from primary explosives Biofuels Uses for surplus glycerol New hard materials to reduce wear Electronic waste Smart growth The book covers traditional green chemistry topics including catalysis benign solvents and alternative feedstocks It also discusses relevant but less frequently

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