Mark Scheme

QU.

Question	Acceptable Answer	Additional Guidance	Mark
(a) (i)	$CaCO_2(s) + 2HCl(eq) \rightarrow CaCl_2(eq) + H_2O(t) + CO_2(g)$	Accept $CaCO_5(s) + 2H^*(aq) \rightarrow Ca^{2^*}(aq) + H_5O(l) + CO_5(g)$	(5)
	Balanced equation (1)		
	State symbols (1)	2nd mark dependent on first or near miss.	
		Reject HyCO ₂ (aq) in equation, but allow state symbol mark if otherwise correct.	

Question	Acceptable Answer	Additional Guidance	Mark
(a) (iii)	Finds molar mass of calcium carbonate	Example of calculation Mr of calculation carbonate = 40.1 + 12 + (16 x 3) = 100.1 (g mof ⁴) Allow = 40 + 12 + (16 x 3) = 100 (g mof ⁴) Accept answer with no working	(x)

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13+ Science Chemistry Mark Scheme (Summer Term 2012) Iseb Iseb,2012-06-08 Edexcel AS/A2 Chemistry Student Unit Guide: Units 3 and 6 Chemistry Laboratory Skills George Facer,2010-02-26 Student Unit Guides are perfect for revision Each guide is written by an examiner and explains the unit requirements This guide offers advice on preparing for the Edexcel Unit 3 and Unit 6 Chemistry Laboratory Skills assessments 13+ Science Chemistry Mark Scheme (Spring Term 2012) Iseb Iseb,2012-01-12 13+ Science Chemistry Mark Scheme (Summer Term 2009) Iseb Iseb,2009-06-08 13+ Science Chemistry Mark Scheme (Spring Term 2009) Iseb Iseb,2009-02-08 13+ Science Chemistry Mark Scheme (Summer Term 2011) Iseb Iseb,2011-06-15

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