Ideal Gas Law: Multiple Choice

The volume of 2.357 mol of oxygen gas at STP is:
 A) 0.1052 L
 B) 52.80 L
 C) 9.504 L
 D) 37.71

ANS: B

1. What is the volume, in liters, occupied by 1.73 moles of $N_{\rm f}$ gas at 0.992 atm pressure and a temperature of 75 °C?

A) 10.7 L B) 33.8 L C) 49.8 L D) 52.2 L

ANS: C

 A sample of 8 grams of oxygen is at STP (1.01 x 10⁵ Pa and 0 °C). How many molecules are present in the sample?

A) 1.51 x 10²³ molecules B) 4.82 x 10²⁴ molecules C) 3.01 x 10²³ molecules D) 6.02 x 10²⁴ molecules E) none of the above

ANS: A

 A certain mass of helium gas occupies 230 mL at 7.5 °C and 2.3 x 10° Pa. How many moles are present in the sample?

A) 0.8484 moles B) 0.0103 moles C) 0.0039 moles

D) 0.0227 moles E) none of the above

ANS: D

What is the number of moles in 500 L of He gas at STP?
 0.05 mol
 20.2 mol
 22 mol
 90 mol

ANS: C

What is the number of moles in 500 L of He gas at SATP?
 A) 20.2 mol
 B) 0.2 mol
 C) 22 mol
 D) 40 mol

ANS: A

ANS: B

Calculate the volume of 12.0 g of helium at 100 °C and 1.2 atm.
 A) 0.013 L
 B) 76.5 L
 C) 7.750 L
 D) 56.0 L

Ideal Gas Law Multiple Choice Question

ML Yell

Ideal Gas Law Multiple Choice Question:

Unveiling the Energy of Verbal Art: An Emotional Sojourn through Ideal Gas Law Multiple Choice Question

In some sort of inundated with displays and the cacophony of quick interaction, the profound energy and mental resonance of verbal art usually disappear into obscurity, eclipsed by the continuous barrage of sound and distractions. Yet, situated within the lyrical pages of **Ideal Gas Law Multiple Choice Question**, a interesting work of literary elegance that pulses with organic feelings, lies an unforgettable trip waiting to be embarked upon. Written with a virtuoso wordsmith, this exciting opus courses visitors on an emotional odyssey, lightly revealing the latent possible and profound impact embedded within the elaborate web of language. Within the heart-wrenching expanse of this evocative examination, we shall embark upon an introspective exploration of the book is key themes, dissect their fascinating writing design, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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