

Name: _____

Date: _____

Student Exploration: Ideal Gas Law

Vocabulary: atmosphere, Avogadro's law, Boyle's law, Charles's law, dependent variable, directly proportional, Gay-Lussac's law, ideal gas, ideal gas constant, ideal gas law, independent variable, inversely proportional, Kelvin temperature scale, kilopascal, mole, pressure, proportionality, STP, volume

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

- Why is it often necessary to add air to your car tires during the winter? Air expands when heated and contracts when cooled – as ambient temperatures get colder, the tires' inflation pressure is going down.

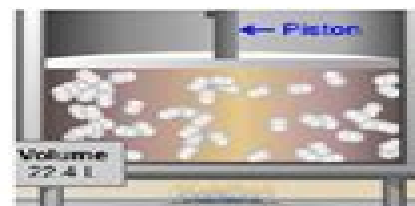


- Why do you think it might be a bad idea to throw an aerosol can into a fire?

Throw an aerosol can into a fire will boil all the liquid contents into gases, which will at that point be highly compressed. Even if the valve ruptures and the contents begin venting, the internal pressure will rapidly build to the point that the can will rupture explosively.

Gizmo Warm-up

The *Ideal Gas Law* Gizmo shows molecules moving within a chamber fitted with a movable piston. As the piston moves up and down, the **volume** of the chamber changes. Since gases expand to fill their container, any changes in the volume of the chamber changes the volume of the gas within.



- Next to **Dependent variable**, check that **Volume** is selected. Using the green slider, change the **pressure**. Note what happens to the temperature, volume, and amount of gas.

What changes? Volume What stays the same? Temperature temperature and amount of gas

- Using the purple slider on the tank of gas, adjust the number of **moles**, or amount of gas.

What changes? volume What stays the same? Pressure and temperature

Ideal Gas Law And Answer Key

David Halliday, Robert Resnick, Jearl Walker



Ideal Gas Law And Answer Key:

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Table of Contents Ideal Gas Law And Answer Key

1. Understanding the eBook Ideal Gas Law And Answer Key
 - The Rise of Digital Reading Ideal Gas Law And Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Ideal Gas Law And Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ideal Gas Law And Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ideal Gas Law And Answer Key
 - Personalized Recommendations
 - Ideal Gas Law And Answer Key User Reviews and Ratings
 - Ideal Gas Law And Answer Key and Bestseller Lists
5. Accessing Ideal Gas Law And Answer Key Free and Paid eBooks
 - Ideal Gas Law And Answer Key Public Domain eBooks
 - Ideal Gas Law And Answer Key eBook Subscription Services
 - Ideal Gas Law And Answer Key Budget-Friendly Options
6. Navigating Ideal Gas Law And Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Ideal Gas Law And Answer Key Compatibility with Devices
 - Ideal Gas Law And Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ideal Gas Law And Answer Key
 - Highlighting and Note-Taking Ideal Gas Law And Answer Key
 - Interactive Elements Ideal Gas Law And Answer Key
8. Staying Engaged with Ideal Gas Law And Answer Key

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ideal Gas Law And Answer Key
9. Balancing eBooks and Physical Books Ideal Gas Law And Answer Key
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ideal Gas Law And Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ideal Gas Law And Answer Key
- Setting Reading Goals Ideal Gas Law And Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ideal Gas Law And Answer Key
- Fact-Checking eBook Content of Ideal Gas Law And Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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