

ICP First Semester Test Review

Chapter 1

1. Steps of the scientific method:
 - a. Form a hypothesis
 - b. Plan experiment
 - c. Collect data
 - d. Analyze
 - e. Conclusion/Modify
2. What is the difference between a scientific theory and a scientific law?
 - a. Theory- an idea tested repeatedly to get the same results
 - b. Law- explains things, but does not describe them — no "why" factor
3. Why is it important to have a control in an experiment?
 - a. The control gives a clear comparison object. Let's you know what would happen if none of the circumstances were applied.
4. Jimmy is conducting an experiment to determine which melts faster on his driveway, sodium chloride or magnesium chloride. In his experiment, what is the dependent variable and what is the independent variable?
 - a. Independent- item used (sodium or magnesium)
 - b. Dependent- time
5. How are scientific models important? Give an example.
 - a. Models are important because they can show you the scale size and objects in an object that is either too small or too large for a human to easily study.
6. Why do scientists use the SI system of measurements?
 - a. It is a world-wide measurement system that translates to every scientist. Information is more easily shared this way.
7. What is the SI base unit for:
 - a. Mass - gram
 - b. Volume - liter
 - c. Time - seconds
 - d. Temperature - Kelvin
8. Convert 658 mg into kg
 - a. .000658 kg
9. Convert 23 km into mm
 - a. 230,000,000 mm
10. Explain the difference between weight and mass.
 - a. Weight is the measurement of how much pressure from gravity something is putting on the ground.
 - b. Mass is the amount of matter in an object.
11. Write the following numbers in scientific notation:
 - a. $12345 = 1.2345 \times 10^4$
 - b. $0.0555 = 5.55 \times 10^{-2}$
12. Explain the difference between precision and accuracy. How can a series of measurements be precise, but not accurate?

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