- 1. Round your answers to the following to the correct # of sig figs.
  - a) 9.767 + 3.45 + 0.73 =
  - b) (4.0)(3170) =
  - c) (456.0)/(1.8) =
- 2. 192.°C = \_\_\_\_\_ k 63.6 k = \_\_\_\_\_ °C
- The boiling point of liquid helium is -269°C. What is this temperature in kelvin?
- 4. The temperature required to melt NaCl is 801 K. What is this temperature in Celsius?
- 5. Which of the following is the largest volume?
- a. 1.3 🗆 107 pL
- b. 1.5 □ 108 nL
- c. 5.0 □ 105 mL
- d. 7.0 102 mL
- e. 3.0 0 101 L
- Wirtz, late for work, gets busted on Rt 302 doing 67 miles/hr. Convert this speed to feet/min.
- A rectangular box has dimensions of 20.0 cm x 15.0 cm x 8.00 cm. Calculate the volume of the box in liters.
- If the fuel efficiency of an automobile is 22 miles per gallon, what is its fuel efficiency in kilometers per liter? (1 km = 0.621 mile, 1.000 L = 1.057 quarts, 4 quarts = 1 gallon)

# **Honors Chemistry Gas Laws Review**

**B** Lingard

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