

### balance nuclear reaction equations

aluminum-26 undergoes positron emission  
potassium-40 undergoes beta decay (electron)  
uranium-238 undergoes alpha decay



### electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



### NOMENCLATURE [naming compounds, writing formulas]

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride  $\text{SF}_6$
- b. Lithium nitride  $\text{Li}_3\text{N}$
- c. Chromium (III) carbonate  $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride  $\text{SnCl}_2$
- e. Ammonium acetate  $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$  (or  $\text{CH}_3\text{COONH}_4$ )
- f. Mercury (I) chloride  $\text{Hg}_2\text{Cl}_2$
- g. Potassium bromate  $\text{KBrO}_3$
- h. Hydrosulfuric acid  $\text{H}_2\text{S(aq)}$
- i. Chloric acid  $\text{HClO}_3$
- j. Sulfurous acid  $\text{H}_2\text{SO}_3$

Name each of the following compounds.

- a.  $\text{CuSO}_4$  Copper (II) sulfate
- b.  $\text{AlF}_3$  aluminum fluoride
- c.  $\text{HI}$  hydrogen iodide
- d.  $\text{NO}$  nitrogen monoxide
- e.  $\text{H}_2\text{Se(aq)}$  hydroselenic acid
- f.  $\text{HNO}_3$  nitric acid
- g.  $\text{HNO}_2$  nitrous acid
- h.  $\text{NaHSO}_4$  sodium hydrogen sulfate

### Half-life

Tritium ( $\text{H}-3$ ) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} 12.5 \text{ mg}$$

### Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 & \times & .692 = 43.596 \rightarrow 44 \\ 65 & \times & .308 = 20.02 \rightarrow 20. \\ & + & \\ & \hline & & 63.62 \rightarrow 64 \end{array}$$

# Honors Chemistry Exam Study Guide Answers

**Princeton Review**



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*Student-Centered Learning by Design* Jacquelyn Whiting, 2021-03-15 Current research is pushing schools to adopt more student centered approaches to the classroom experience and educators librarians and classroom teachers alike are being challenged to revise their curricula and instruction to be student centered personalized and differentiated This book empowers librarians teachers and administrators to be empathic problem solvers and decision makers By reframing the challenges that members of a learning community face as opportunities to better meet teaching and learning needs readers will find that adoption of a mindset focused on users namely design thinking elevates and creates opportunities for innovating pedagogy Moreover it can enhance school culture as well as build channels of communication among various stakeholders in schools and districts When educators of any subject or discipline apply design thinking skills to their curriculum implementation authentic student centered learning experiences become the core of the learning experience The case studies shared in this book provide examples of student centered approaches being used in

elementary middle and high schools so that readers have many models on which to base their work and from which to build confidence in shifting their pedagogy to keep the student at the center of teaching and learning decisions **Chemistry All-in-One For Dummies (+ Chapter Quizzes Online)** Christopher R. Hren, John T. Moore, Peter J. Mikulecky, 2022-11-23 Everything you need to crush chemistry with confidence Chemistry All in One For Dummies arms you with all the no nonsense how to content you'll need to pass your chemistry class with flying colors You'll find tons of practical examples and practice problems and you'll get access to an online quiz for every chapter Reinforce the concepts you learn in the classroom and beef up your understanding of all the chemistry topics covered in the standard curriculum Prepping for the AP Chemistry exam Dummies has your back with plenty of review before test day With clear definitions concise explanations and plenty of helpful information on everything from matter and molecules to moles and measurements Chemistry All in One For Dummies is a one stop resource for chem students of all valences Review all the topics covered in a full year high school chemistry course or one semester of college chemistry Understand atoms molecules and the periodic table of elements Master chemical equations solutions and states of matter Complete practice problems and end of chapter quizzes online Chemistry All In One For Dummies is perfect for students who need help with coursework or want to cram extra hard to ace that chem test

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research based approach to teaching thinking begun at Harvard's Project Zero that develops students' thinking dispositions while at the same time deepening their understanding of the topics they study. Rather than a set of fixed lessons, Visible Thinking is a varied collection of practices including thinking routines, small sets of questions or a short sequence of steps as well as the documentation of student thinking. Using this process, thinking becomes visible as the students' different viewpoints are expressed, documented, discussed, and reflected upon. Helps direct student thinking and structure classroom discussion. Can be applied with students at all grade levels and in all content areas. Includes easy-to-implement classroom strategies. The book also comes with a DVD of video clips featuring Visible Thinking in practice in different classrooms.

**Days of Honor and Honesty** Mary Ita Malone, 2018-07-25 While growing up in Carlow, Ireland, Mary Ita Malone's early dream to become a missionary in Africa led her to join a religious order. But as she was about to discover, life has a way of surprising all of us when we least expect it. After Malone attended medical school in Dublin, she was assigned to Ortu Mission Hospital in West Pokot, Kenya, to bring medical services to the neglected indigenous people. Driven by her desire to provide preventive rather than curative care, Malone eventually earned a master's degree in public health in America. But when her dreams of returning to Ortu to continue her work were dashed, Malone reveals how she worked through her internal struggles to teach at the medical school in Nairobi and eventually immigrate to the United States to specialize in physical medicine and rehabilitation. In this fascinating memoir, a missionary nun and physician shares the true story of her tumultuous and exotic journey as she followed her guiding star from a Catholic religious order in Ireland to the far reaches of Africa and finally to America.

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