## Name:

## Honors Chem Test: Ch. 7

	1	How many atoms of fluorine are present in a molecule of carbon tetrafluoride, CFV			
		a. 1	e.	4	
		b. 2	d.	5	
	2.	Name the compound Al <sub>2</sub> S <sub>3</sub>			
		a. aluminum sulfate	- C	aluminum(II) sulfate	
		<ul> <li>aluminum sulfur</li> </ul>	d.	aluminum sulfide	
	3.	Name the compound CF.			
		a. cal crum fluoride	c.	carbon tetrafiuon de	
		<ul> <li>b. carbon fluoride</li> </ul>	d.	monocalcium quadrafluoride	
	4.	Name the compound $N_2O_3$ .			
		a. dimitrogen oonde	C	mitric coulde	
		b. nitrogen triconde	d.	dinitrogen triconde	
_	5.	What is the formula for deastsenic trisulfide?			
		a. Ar <sub>2</sub> S <sub>3</sub>	· c	As <sub>2</sub> S <sub>4</sub>	
		b. ArSı	d.	A53S3	
	6.	What is the formula for sulfur dichloride?			
		a. SCI	0.00	S <sub>2</sub> CI	
		b. SCI;	d.	S,CI,	
	7.	What is the formula for hydrochloric ac	id?		
		a. HF	, c.	HCIO	
		ь нсі	d.	H <sub>2</sub> CO <sub>2</sub>	
	8.	The molar mass of an element is the mass of one			
		a. atom of the element.	c.	gram of the element.	
		<ul> <li>b. crystal of the element.</li> </ul>	d.	mole of the element.	
-	9.	The molar mass of MgI <sub>2</sub> is			
		a. the sum of the masses of 1 mol of Mg and 2 mol of 1.			
		<ul> <li>the sum of the masses of 1 mol of Mg and 1 mol of I.</li> </ul>			
		c. the sum of the masses of 1 atom of			

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