

2012/2013 AP Biology Midterm Review Sheet

- **In-Class Midterm Date: Wednesday, January 24, 2013 – in the LIBRARY – 11:45-1:45**
- **Format:** 45-50 Multiple Choice Questions, two essays (one lab related, one content related) – 2 hours
 - Be prepared to work at this fast pace... which is actually slower than the AP Exam pace!!
 - Some questions will require calculations (as on the AP test)
- **Bring:** # 2 pencils & a blue or black pen, 4-function calculator. No reference materials may be used.
- The midterm will count under Criterion C.
- Use this review sheet to help guide you during your preparation for this exam.
- Use prior quizzes and tests, as well as in class activities and labs to help you review

Unit 1: Chemistry of Life

Chapter 3 – Water

- What types of bonds do you find within a water molecule? Between water molecules?
- What atom in water is very electronegative?
- Properties: excellent solvent, high heat capacity—moderating influence, evaporative cooling, ice floats, strong cohesion & surface tension, strong adhesion
- pH scale 0-14 acid = 0-(7) high $[H^+]$, low $[OH^-]$ neutral = 7 base (7)-14, low $[H^+]$, high $[OH^-]$
- Buffers
- Water Lab

Chapter 4/5 – Carbon / Organic Molecules

- What are some properties of the carbon (organic) atom?
- Functional Groups: structural formulas, properties, and examples of molecules they are found in:
 - Amino, central carbonyl, terminal carbonyl, carboxyl, hydroxyl, methyl, phosphate, sulfhydryl
- What is the difference between dehydration/condensation synthesis and hydrolysis?
 - $C_6H_{12}O_6 + C_6H_{12}O_6 \rightarrow C_{12}H_{22}O_{11} + H_2O$ (disaccharide) + H_2O
- Know macromolecule monomers vs. polymers, bonds, and examples:
 - Carbohydrates: monosaccharides, disaccharides, polysaccharides, sugars, starch, glycogen, cellulose, chitin, glycosidic linkage bonds
 - Proteins - structure, transport, defense, enzymes, amino acids, dipeptides, polypeptides, proteins, peptide bonds, 1°, 2°, 3°, 4° levels of structure
 - Lipids - energy storage, structure, hormones, groups: triglycerides (fats, saturated C-C, unsaturated/kinky C=C), phospholipids, steroids, (cholesterol, sex hormones), ester linkage bonds
 - Nucleic acids- genetic information storage, structure: nucleotides (pentose sugar, nitrogenous base (A,T,C,G,U), negatively charged phosphate groups, DNA, RNA, hydrogen bonds between bases, covalent bonds in backbone
- Organic molecules lab

Chapter 8 - Enzymes/ Metabolism

- What do the first and second laws of thermodynamics state?
- What is the structure and purpose of ATP?
- What happens when high-energy phosphate bonds are broken in ATP?
- $\Delta G_{\text{reaction}}$ (Change in Energy of Reaction) = G_{products} (Energy of Products) – $G_{\text{reactants}}$ (Energy of Reactants)
- Define: Catabolism, Hydrolysis, Exergonic, Negative Delta G
- What macromolecule are enzymes made of? What do they do to a substrate? Can they be reused? What do enzymes do to the activation energy of a reaction?
- Function: metabolic catalysts, lock & key model, induced fit model, "-ase", substrate specific
- Factors affecting enzyme action: pH, temperature, coenzymes, cofactors, substrate concentration, enzyme concentration
- Activators (ex: allosteric) vs. inhibitors (ex: competitive or non-competitive)
- Allosteric site vs. active site
- Negative feedback loop / feedback inhibition
- Enzyme Lab- Catecholase – What factors affect the activity of catecholase?

Honors Biology Midterm Review Answers 2013

Robert R. Redfield



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Table of Contents Honors Biology Midterm Review Answers 2013

1. Understanding the eBook Honors Biology Midterm Review Answers 2013
 - The Rise of Digital Reading Honors Biology Midterm Review Answers 2013
 - Advantages of eBooks Over Traditional Books
2. Identifying Honors Biology Midterm Review Answers 2013
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Honors Biology Midterm Review Answers 2013
 - User-Friendly Interface
4. Exploring eBook Recommendations from Honors Biology Midterm Review Answers 2013
 - Personalized Recommendations
 - Honors Biology Midterm Review Answers 2013 User Reviews and Ratings

- Honors Biology Midterm Review Answers 2013 and Bestseller Lists
- 5. Accessing Honors Biology Midterm Review Answers 2013 Free and Paid eBooks
 - Honors Biology Midterm Review Answers 2013 Public Domain eBooks
 - Honors Biology Midterm Review Answers 2013 eBook Subscription Services
 - Honors Biology Midterm Review Answers 2013 Budget-Friendly Options
- 6. Navigating Honors Biology Midterm Review Answers 2013 eBook Formats
 - ePub, PDF, MOBI, and More
 - Honors Biology Midterm Review Answers 2013 Compatibility with Devices
 - Honors Biology Midterm Review Answers 2013 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Honors Biology Midterm Review Answers 2013
 - Highlighting and Note-Taking Honors Biology Midterm Review Answers 2013
 - Interactive Elements Honors Biology Midterm Review Answers 2013
- 8. Staying Engaged with Honors Biology Midterm Review Answers 2013
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Honors Biology Midterm Review Answers 2013
- 9. Balancing eBooks and Physical Books Honors Biology Midterm Review Answers 2013
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Honors Biology Midterm Review Answers 2013
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Honors Biology Midterm Review Answers 2013
 - Setting Reading Goals Honors Biology Midterm Review Answers 2013
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Honors Biology Midterm Review Answers 2013
 - Fact-Checking eBook Content of Honors Biology Midterm Review Answers 2013
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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