

Student directions *Beta Decay*

This is written to follow Loeblein's activity for *Alpha Decay*
<https://phet.colorado.edu>

Learning Goals: Students will be able to:

- Describe the process of Beta decay
- Differentiate between Alpha and Beta decay
- Compare the meaning of "Half-life" for Alpha and Beta decay.

Directions: Open *Beta Decay*

1. Investigating "Beta Decay"

- Start on the Single Atom tab - observe the decay of Hydrogen-3 and Carbon- 14. Use Reset Nucleus to watch the process repeatedly. Write a description of what happens in the beta decay of an atom.
- Check your ideas with the "Custom" atom and reflect on your ideas.
New ideas here;
- Verify your ideas by using online resources to determine what the differences are between Hydrogen-3 and Helium-3 as well as Carbon-14 and Nitrogen-14. Also, use other resources to see what "Beta Decay" means and cite at least one valid source.
Cites here;
- Practice using your ideas by predicting what would happen if the following undergo beta decay:
 - Carbon -10 $\rightarrow$ _____ + _____
 - Cesium-137 $\rightarrow$ _____ + _____
 - Thorium-234 $\rightarrow$ _____ + _____
- Practice using your ideas by predicting what would happen. Uranium-238 undergoes alpha decay and then beta decay.
Uranium-238 $\rightarrow$ _____

2. Investigating "Half-life" for Beta Decay

- Use the Charts at the top of the sim to test ideas you might have about half-life. Make sure to use multiple samples and substances with a variety of half-lives. Make a data table that shows your tests.
Data Table here;

Holt Science Beta Decay Sim Answers

H Kauffman



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Nuclear Science Abstracts ,1975-04 Science Citation Index ,1995 Vols for 1964 have guides and journal lists *INIS*
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