

Naming Chemical Compounds Worksheet

Name the following *ionic* compounds:

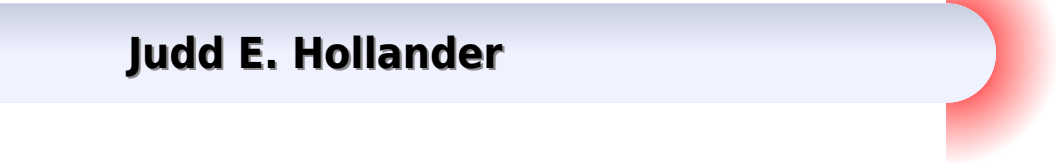
- | | | |
|-----|--|------------------------|
| 1) | NaBr | <u>Sodium Bromide</u> |
| 2) | CaO | <u>Calcium Oxide</u> |
| 3) | Li ₂ S | <u>Lithium Sulfide</u> |
| 4) | MgBr ₂ | _____ |
| 5) | Pu(OH) ₃ | _____ |
| 6) | Hg(CN) ₂ | _____ |
| 7) | Mo(C ₂ H ₃ O ₂) ₂ | _____ |
| 8) | Cr(CrO ₄) ₃ | _____ |
| 9) | WO ₂ | _____ |
| 10) | Mg(ClO) ₂ | _____ |

Write the formulas for the following *ionic* compounds:

- | | | |
|-----|--------------------------------|-------|
| 11. | potassium dichromate | _____ |
| 12. | gold(III) oxide | _____ |
| 13. | aluminum periodate | _____ |
| 14. | barium nitrite | _____ |
| 15. | silver carbonate | _____ |
| 16. | lithium sulfite | _____ |
| 17. | zinc hydrogen carbonate | _____ |
| 18. | iron(III) hydroxide | _____ |
| 19. | ammonium phosphate | _____ |
| 20. | copper(II) bromite | _____ |
| 21) | SO ₃ | _____ |
| 22) | N ₂ S | _____ |
| 23) | BF ₃ | _____ |
| 24) | P ₂ Br ₄ | _____ |
| 25) | SiO ₂ | _____ |
| 26) | SF ₆ | _____ |
| 27) | NO ₂ | _____ |
| 28) | nitrogen trichloride | _____ |
| 29) | dinitrogen trioxide | _____ |
| 30) | phosphorus pentafluoride | _____ |
| 31) | diboron tetrahydride | _____ |

Holt Unit 7 Bonding Naming And Chemical Reactions Test

Judd E. Hollander



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