

# ENGLISH LANGUAGE TEST — GRADE NINE

## Mock Exam 2010-2011

School: \_\_\_\_\_

Class: \_\_\_\_\_

Student: \_\_\_\_\_

TIME: 2 HOURS

TOTAL: 30 MARKS

### LISTENING 1

You are going to hear five people speaking. What are they doing? For each speaker, choose ONE of the items in the box. Write **A, B, C**, etc, as in the example.

Example: "Now students. Who can tell me the correct answer to number five?" F

1. \_\_\_\_\_

2. \_\_\_\_\_

3. \_\_\_\_\_

4. \_\_\_\_\_

5. \_\_\_\_\_

### ACTIVITIES

- A cleaning the house
- B watching television
- C playing football
- D driving a car
- E doing homework
- F teaching a class
- G cooking food
- H talking on the phone

### LISTENING 2

You are going to hear a text about pottery. Listen, and for each item, choose the correct option: **A, B** or **C**. Write your answer in the box on the right.

- |  |                |              |
|--|----------------|--------------|
| 6. Pottery is made of _____                    |                |              |
| A. steel                                       | B. clay        | C. wood      |
| 7. Today pottery is painted when it is _____   |                |              |
| A. thin  | B. soft        | C. hard      |
| 8. People who make pottery are called _____    |                |              |
| A. potters                                     | B. carpenters  | C. jewelers  |
| 9. Different _____ are used to paint pottery.  |                |              |
| A. brushes                                     | B. colours     | C. hands     |
| 10. The purest type of pottery is called _____ |                |              |
| A. porcelain                                   | B. earthenware | C. stoneware |


# Grade 9 Final Exam2014

**Michal Rosen-Zvi**



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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g}) \quad \Delta h = 94.5 \text{ kJ}$   
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \quad \Delta h = 71.2 \text{ kJ}$   
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g}) \quad \Delta h = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 2 \text{H}_2\text{O}(\text{l}) \quad \Delta h = 1411 \text{ kJ/mole}$   
 $\text{C}_2\text{H}_6(\text{g}) + 7 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{l}) \quad \Delta h = 1560 \text{ kJ/mole}$   
 $\text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta h = 285 \text{ kJ/mole}$

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web thus for reaction 9 2 in which 1 mole of  $\text{H}_3\text{O}^+$  and 1 mole of  $\text{OH}^-$  combine to form 2 moles of  $\text{H}_2\text{O}$   $\Delta h = 55.8 \text{ kJ}$  as shown note that the enthalpy change  $\Delta h_{\text{products}} - \Delta h_{\text{reactants}}$  is positive if heat is absorbed that is if  $\Delta h_{\text{products}}$

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