(Question 1 continued)

(d) The cross-sectional area of the tube is 1.3 × 10⁻³ m² and the temperature of air is 300 K. Estimate the number of moles of air in the tube.

[2]

[2]

(e) The equation in (b) may be used to predict the pressure of the air at extremely large values of ¹/_H. Suggest why this will be an unreliable estimate of the pressure.

Very large 1/H means small volume. Small volume means high cressure.

In ideal gas law, it stated that molecules are not close to each other and hence the forces between them are negligible.

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