
KAPLAN DIAGNOSTIC A ACTUAL EXAM: COMPLETE EXAM WITH CORRECT VERIFIED ANSWERS | COMPREHENSIVE ASSESSMENT FOR ACADEMIC READINESS AND PREPARATION| GRADED A

1. The nurse cares for the client diagnosed as being in the manic phase of bipolar disorder. Which behavior indicates to the nurse the client condition is improving?

- a. The client offers suggestions to other clients on the unit
- b. The client begins to write a book about life
- c. The client sits and eats with other clients on unit.
- d. The client talks with other clients a group meeting

C

2. The health care provider orders a continuous intravenous aminophylline infusion for a two year old client. It is most important for the nurse to intervene for which situation?

- a. The client heart rate is 100 bpm
- b. The clients blood pressure is 100/60 mmHg
- c. The clients serum theophylline level is 25 mcg/mL.
- d. The client is sleepy

C?

3. The nurse teaches the client about the schedule cardiac catheterization. Which statement, if made by the client to the nurse, indicates that the teaching was effective?

- a. "I understand that there is little or no risk associated with this procedure."
- b. "I may experience a little pounding sensation in my chest during the procedure."
- c. "I will be in and out of the procedure room in about 30 minutes."
- d. "I will be able to walk in the hall soon after the procedure is completed."

4. During the second stage of labor, the client's partner asks the nurse, "Can I go get a cup of coffee from the cafeteria?" Which response by the nurse is best?

- a. "I will get you a cup of coffee."
- b. "It would be best if you stayed here at this time."
- c. "Ask your partner if it is acceptable to leave."
- d. "Why do you want to leave the room?"

Kaplan University Academic Readiness Assessment

Todd Kettler, Cheryl Taliaferro



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For Profit Higher Education ,2012 The Great American Education-Industrial Complex Anthony G. Picciano,Joel Spring,2013-05-07 The Great American Education Industrial Complex examines the structure and nature of national networks and enterprises that seek to influence public education policy in accord with their own goals and objectives In the past twenty years significant changes have taken place in the way various interest groups seek to influence policies and practices in public education in the United States No longer left to the experience and knowledge of educators American education has become as much the domain of private organizations corporate entities and political agents who see it as a market for their ideas technologies and ultimately profits Piccciano and Spring posit that educational technology is the vehicle whereby these separate movements organizations and individuals have become integrated in a powerful common entity and detail how the educational industrial complex has grown and strengthened its position of influence This timely carefully documented well argued book brings together Picciano s perspective and expertise in the field of technology and policy issues and Spring s in the history and politics of education in a unique critical analysis of the education industrial complex and its implications for the future **Handbook on Measurement, Assessment, and Evaluation in Higher Education** Charles Secolsky,D. Brian Denison,2017-07-31 In this valuable resource well known scholars present a detailed understanding of contemporary theories and practices in the fields of measurement assessment and evaluation with guidance on how to apply these ideas for the benefit of students and institutions Bringing together terminology analytical perspectives and methodological advances this second edition facilitates informed decision making while connecting the latest thinking in these methodological areas with actual practice in higher education This research handbook provides higher education administrators student affairs personnel institutional researchers and faculty with an integrated volume of theory method and application **Personalized Learning in Gifted Education** Todd Kettler,Cheryl Taliaferro,2022-02-16 Gifted students can exhibit extreme variance in both their abilities and their interests yet they are often treated within schools as one homogeneous specialized population Personalized Learning in Gifted Education helps educators strengthen their differentiation of both instruction and services for advanced students This book helps educators develop the specific gifts and talents of the gifted students they serve demonstrates how educators can utilize the wealth of data they have at their disposal provides a rationale and blueprint for a stronger more personalized approach to gifted education offers suggestions for both elementary and secondary schools Recommendations center around five features of personalized learning personalized learning plans project or problem based learning competency based progression through the curriculum criterion referenced assessments and multi year mentoring TOEFL Paper-and-Pencil Janet Shanks,2004 Are you ready for the TOEFL R Exam You will be Kaplan s TOEFL R Paper and Pencil with 3 Audio CDs Third Edition is a complete review of all the material on the paper based and computer based TOEFL R exam You get 3 realistic practice tests Kaplan s powerful test taking strategies

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the field of gifted education related to the areas of equitable identification implementation of services and programmatic assessment These protocols seek to initiate discussion and critical discourse regarding diverse gifted learners among higher education faculty state department personnel district administrators and classroom teachers Covering topics such as digital differentiation equitable assessment and STEM education this text is ideal for teacher education programs preparation programs university degree programs university credential programs certificate programs faculty graduate students state departments of education superintendents coordinators administrators teachers professors academicians and researchers

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the objectives of the boiling point determination lab were to determine the boiling point of a mystery substance provided to each student boiling point is the point at which a liquid changes physical states and becomes a vapor or gas boiling point is important to understand because it can be used to help distinguish between different substances

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at agiven pressure the boiling

point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube end c

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boiling point determination lab report conclusion objectivesto correctly measure boiling issues of various liquids and identify an unknown liquid based on its boiling point background information objekte can be distinguished into acid properties and physical properties

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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introduction in this lab we will be using boiling points melting points and infrared spectrometry to identify unknown compounds boiling points are affect by three key factors the first being intermolecular forces inter molecular affect boiling point based on bond strength

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it

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clean up solid and liquid spills immediately true g conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b

had a melting point range of 122.6–124.5 °C thus

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Nov 27 2019 essay sample the purpose of this experiment is to determine the boiling points of various organic compounds and to use these to identify unknowns equipment free essays

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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Aug 11 2020 boiling points are very sensitive to changes in applied pressure so all boiling points should be reported with the measured pressure a compound's normal boiling point refers to its boiling point at a pressure of 760 mm Hg

experiment 11 vapour pressure proctech 2ce3 lab - Jul 07 2022

the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine the vapour pressure of a pure liquid at various temperatures

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exploring boiling points abstract in this experiment we are given a set of straight chain alkanes and a set of compounds to compare by following the trendline of the straight chain alkanes it could be concluded that as the molar mass of the straight chain alkanes increases their boiling points increase as well

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laboratory partners samantha and mackenzie goals □ correctly determine the boiling point of an unknown liquid □ measure the refractive index of an unknown □ use the gas chromatogram to determine the mass percentage of a mixture and

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50 °C so in order to convert an experimental measurement taken in Calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that need to be determined by physical properties such as boiling point melting point density solubility etc in this experiment i am to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a bunsen burner boiling

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there are a variety of methods by which a sample s boiling point can be determined including distillation reflux and by using a thiele tube the most straightforward method uses a thiele tube and has the advantage of using less than 0 5ml 0 5 ml of material