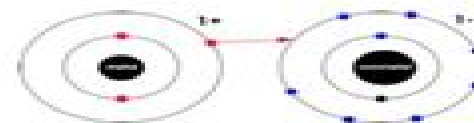


## DRAWING IONIC BONDS

### Introduction:

An ionic bond forms between a metal and a nonmetal. An ionic bond occurs when a metal atom **loses valence electron(s)** to become a positive (+) ion or **cation**, while the nonmetal **gains valence electron(s)** to become a negative (-) ion or **anion**. An ionic bond forms due to the strong electrostatic attraction between the **positive and negative ions**.



### How to Draw Simple Ionic Bond Diagrams

1. Draw the electron dot diagrams for each element.

***TIP:** This will help you see the number of unpaired valence electrons for each element, which will help you determine how many electrons each element must gain or lose to achieve a full outer shell.*

2. Draw arrows to show the transfer of valence electrons from the metal to the nonmetal.

# WORKSHEET



## Drawing Ionic & Covalent Bonds 2 Worksheet Set

|    | Metal                                                   | Nonmetal                                                        | Formula                                    | Bonding Diagram |
|----|---------------------------------------------------------|-----------------------------------------------------------------|--------------------------------------------|-----------------|
| A. | sodium<br>(must lose 1)<br><br>$\text{Na} \cdot$        | chlorine<br>(must gain 1)<br><br>$\cdot \ddot{\text{Cl}} \cdot$ | $\text{NaCl}$<br><br>(shows 1 Na and 1 Cl) |                 |
| B. | calcium<br>(must lose 2)<br><br>$\cdot \text{Ca} \cdot$ | fluorine<br>(must gain 1)<br><br>$\cdot \ddot{\text{F}} \cdot$  | $\text{CaF}_2$<br><br>(shows 1 Ca and 2 F) |                 |

# Ionic Covalent Bond Practice Worksheets

**L Darling-Hammond**



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