

- A.  $\text{H}_2\text{O}(g) + 2\text{CO}(g) \rightarrow \text{heat} + 2\text{H}_2\text{O}(l)(g)$
- B.  $\text{H}_2\text{O}(g) + 2\text{H}_2\text{O}(g) \rightarrow 2\text{H}_2\text{O}(g) + \text{heat}$
- C.  $\text{CH}_4(g) + \text{H}_2\text{O}(g) \rightarrow \text{heat} + \text{CO}(g) + 2\text{H}_2\text{O}(g)$
- D.  $\text{C}_5\text{H}_{12}(g) + 4\text{H}_2\text{O}(g) \rightarrow \text{CH}_4(g) + 2\text{H}_2\text{S}(g) + \text{heat}$

26. Consider the following equilibrium system:  $\text{H}_2(g) + 2\text{H}_2\text{O}(g) \rightleftharpoons 2\text{H}_2\text{O}(g) + 92 \text{ kJ}$   
In order to maximize the yield for this reaction, the best conditions are:

- A. Low pressure and low temperature
- B. Low pressure and high temperature
- C. High temperature and low pressure
- D. High pressure and low temperature

27. Consider the following equilibrium:  $\text{CaCO}_3(s) + 556 \text{ kJ} \rightleftharpoons \text{CaO}(s) + \text{CO}_2(g)$   
The equilibrium will shift to the right:

- A.  $\text{CO}_2$  is added
- B.  $\text{CaCO}_3(s)$  is added
- C. The temperature is increased
- D. The temperature is decreased.

28. Explain the difference between Arrhenius acids and bases and Brønsted-Lowry acids and bases.

29. List three characteristics that acids and bases have in common.

30. To raise the pH of a solution, would you add HCl or NaOH? Why?

31. Describe and give an example of an electrolyte.

32. Find the percent composition of each element in  $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ .

33. Given the balanced equation:  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ , how many grams of  $\text{H}_2\text{O}$  can be produced from 50g of  $\text{O}_2$ ?

34. Draw energy diagrams for the following:  
Reactants 50kJ / Products 20kJ  
Products 30kJ / Products 40kJ  
Activation energy 20kJ / Activation energy 50kJ

35. Given the unbalanced equation below, what mass of sodium chloride can be produced from 23 g of Na and 35g of  $\text{Cl}_2$ ?



36. Explain the difference between saturated and unsaturated carbon compounds.

37. Identify the nuclear forces and explain each.

38. Explain how chemical reactions differ from nuclear reactions.

# Honors Chemistry Final Exams

**United States Air Force Academy**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the United States Air Force Academy text.

## **Honors Chemistry Final Exams:**

Honors Chemistry Final Exams: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous compelling novels enthralling the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the engaging narratives that have captivated audiences this year. Honors Chemistry Final Exams : Colleen Hoover's "It Ends with Us" This touching tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover expertly weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed. Honors Chemistry Final Exams : Taylor Jenkins Reid's "The Seven Husbands of Evelyn Hugo" This captivating historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reid's captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens' "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These bestselling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a masterful and thrilling novel that will keep you speculating until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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