

HALF-LIFE PROBLEMS

Name _____ Block _____

1. An isotope of cesium (cesium-137) has a half-life of 30 years. If 1.0 g of cesium-137 disintegrates over a period of 90 years, how many g of cesium-137 would remain?

$$\frac{90}{30} = 3 \frac{1}{2}$$

$\begin{pmatrix} 1.5 \\ 0.5 \\ 2.5 \\ 1.25 \end{pmatrix} g$

2. Actinium-226 has a half-life of 29 hours. If 100 mg of actinium-226 disintegrates over a period of 58 hours, how many mg of actinium-226 will remain?

58-2-44

100
50
25

3. Sodium-25 was to be used in an experiment, but it took 3.0 minutes to get the sodium from the reactor to the laboratory. If 5.0 mg of sodium-25 was removed from the reactor, how many mg of sodium-25 were placed in the reaction vessel 3.0 minutes later if the half-life of sodium-25 is 60 seconds?

$$\frac{3.0}{1.0} = 3.0$$

$\left\{ \begin{array}{l} 5 \text{ mg} \\ 2.5 \\ 1.25 \\ 0.625 \text{ mg} \end{array} \right.$

4. The half-life of isotope X is 2.0 years. How many years would it take for a 4.0 mg sample of X to decay and have only 0.50 mg of it remain?

$$3 \times 2.0 \text{ yr} = 6 \text{ yr}$$

4.00
2.00
1.00
5.00 } 3 h-L

5. Selenium-83 has a half-life of 25.0 minutes. How many minutes would it take for a 10.0 mg sample to decay and have only 1.25 mg of it remain?

$$3 \times 25 = 75 \text{ mi}$$

1.25 3 h.L

6. The half-life of Po-218 is three minutes. How much of a 2.0 gram sample remains after 15 minutes? Suppose you wanted to buy some of this isotope, and it required half an hour for it reach you. How much should you order if you need to use 0.10 gram of this material?

5. chd

2.3
1.5
2.2
1.25
1.0625

302, 10 h.f.

10 h.l

$\cdot 1 \times 2 \times 2 \times 2 \times 2 = 2 \times 2 \times 2 = 2 \times 2 \times 2$

$102.4g$

Half Life Problems Answer Key

N Noddings



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1. Understanding the eBook Half Life Problems Answer Key
 - The Rise of Digital Reading Half Life Problems Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Half Life Problems Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Half Life Problems Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Half Life Problems Answer Key
 - Personalized Recommendations
 - Half Life Problems Answer Key User Reviews and Ratings
 - Half Life Problems Answer Key and Bestseller Lists
5. Accessing Half Life Problems Answer Key Free and Paid eBooks
 - Half Life Problems Answer Key Public Domain eBooks
 - Half Life Problems Answer Key eBook Subscription Services
 - Half Life Problems Answer Key Budget-Friendly Options
6. Navigating Half Life Problems Answer Key eBook Formats

- ePub, PDF, MOBI, and More
- Half Life Problems Answer Key Compatibility with Devices
- Half Life Problems Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Half Life Problems Answer Key
 - Highlighting and Note-Taking Half Life Problems Answer Key
 - Interactive Elements Half Life Problems Answer Key
- 8. Staying Engaged with Half Life Problems Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Half Life Problems Answer Key
- 9. Balancing eBooks and Physical Books Half Life Problems Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Half Life Problems Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Half Life Problems Answer Key
 - Setting Reading Goals Half Life Problems Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Half Life Problems Answer Key
 - Fact-Checking eBook Content of Half Life Problems Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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