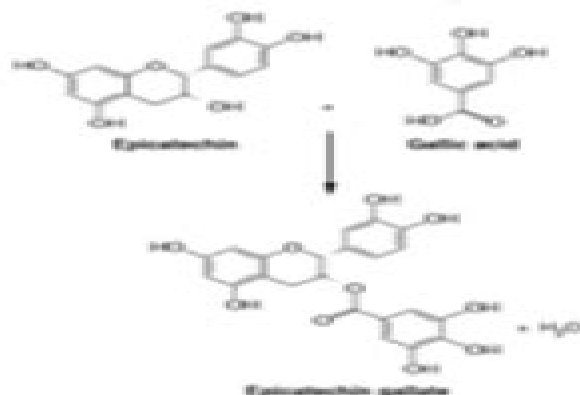


F332 Adv. Notice June 2014 : some "predicted Qs"

Selection of student-suggested Qs based on the advance notice material for F332 June 2014,in no particular order.....

Q1] TCA only needs to be present at about 2 parts per trillion for a bottle of wine to be ruined. What is this as a percentage? (1,000,000 million is a trillion)



Q2] In figure 3 (above), epicatechin and gallic acid react together to form Epicatechin gallate. On the diagram above, circle the carboxylic acid functional group on the Gallic acid molecule. This is known as a "condensation reaction". Suggest why?

Q3] Butanedione, a molecule formed during the aging of Chardonnay wines, is often suspected of being the cause of the buttery taste. Draw the skeletal formula of this molecule.

Q4] Vitamin C (ascorbic acid) is found in the grapes. The molecule contains more than one type of alcohol. On the molecule below, circle and label the different types of alcohol present.



Q5] Oct-1-en-3-ol is a molecule suspected of causing a bad taste in corked wine.



- Suggest a structural isomer of this molecule – draw its structural formula and name it.
- Suggest the reagents and conditions needed to convert oct-1-en-3-ol into oct-1-en-3-one in a chemistry lab.
- Suggest how Infra-Red spectrometry could be used to differentiate between molecules of oct-1-en-3-ol and oct-1-en-3-one.

F332 June 2014 Paper

Rachel Sandford



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