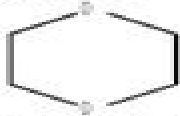


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1. (a) G: CO (1)
 $\text{HCOOH/H}_2\text{SO}_4 \rightarrow \text{CO} + \text{H}_2\text{O (1)}$ 2
- (b) H: C (1)
 $\text{C}_{12}\text{H}_{22}\text{O}_{11} \rightarrow 12\text{C} + 11\text{H}_2\text{O (1)}$ 2
- (c) I: $\text{C}_4\text{H}_8\text{O}_2 \text{ (1)}$
 $2\text{C}_2\text{H}_4\text{O}_2 \rightarrow \text{C}_4\text{H}_8\text{O}_2 + 2\text{H}_2\text{O (1)}$
 Structure:

 (1)
 accept any sensible structure of $\text{C}_4\text{H}_8\text{O}_2$ 3
2. (a) empirical formula
 $\text{N} : \text{O} = 63.64/14 : 36.36/16 \text{ (1)}$
 $= 4.56 : 2.27 = 2 : 1$. Empirical formula = $\text{N}_2\text{O (1)}$
 molecular formula
 M_r of gas = $1.833 \times 24 = 44 \text{ (1)}$ (calc 43.992)
 with these two pieces of evidence, assume that molecular formula = N_2O 3
- (b) any chemical that reacts to produce gas:
 e.g. carbonate and $\text{CO}_2 \text{ (1)}$
 accept: metal more reactive than Pb and H_2
 balanced equation to match chemical added (1) 2
- (c) $M_r(\text{Lidocaine}) = 236 \text{ (1)}$
 moles Novocaine = $100 \times 10^{-3}/236 = 4.24 \times 10^{-4} \text{ (1)}$
 concentration of Novocaine = $4.24 \times 10^{-4} \times (1000/5)$
 $= 0.0847/0.0848 \text{ mol dm}^{-3} \text{ (1)}$ 3
- (d) mass C = $12 \times 3.74/44.0 = 12 \times 0.085 = 1.02 \text{ g (1)}$
 mass H = $2/18 \times 0.918 = 0.102 \text{ g (1)}$
 mass O = $1.394 - (1.020 + 0.102) = 0.272 \text{ g}$
 ratio C : H : O = $1.02/12 : 0.102/1 : 0.272/16 \text{ (1)}$
 $= 0.0850 : 0.102 : 0.0170 / 5 : 6 : 1 / \text{C}_5\text{H}_6\text{O (1)}$
 $\text{C}_5\text{H}_{10}\text{O}$ has relative mass of 82
 M_r is 164 = 2×82
 $\therefore$ molecular formula = $\text{C}_{10}\text{H}_{12}\text{O}_2 \text{ (1)}$ 5
3. (i) $4\text{PCl}_3 + 10\text{MgO} \rightarrow \text{P}_4\text{O}_{10} + 10\text{MgCl}_2 \text{ (1)}$ 1
- (ii) $100\text{g P}_4\text{O}_{10} = \frac{100}{284} / 0.35(2) \text{ mol (1)}$
 moles PCl_3 needed = $4 \times 0.352 = 1.408/1.4 \text{ mol (1)}$
 mass $\text{PCl}_3 = 1.4(08) \times 208.5 = 293.568 / 294 \text{ g/ } 291.9 \text{ g (1)}$

[7]

[13]

F325 Exam Questions Answers Oct

Daniel F McAuley



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OCR A Chemistry A2 Student Unit Guide: Unit F325 New Edition: Equilibria, Energetics and Elements ePub Mike Smith, 2013-02-22 Written by a former senior examiner Mike Smith this OCR A A2 Chemistry Student Unit Guide is the essential study companion for Unit F325 Equilibria Energetics and Elements This full colour book includes all you need to know to prepare for your unit exam clear guidance on the content of the unit with topic summaries knowledge check questions and a quick reference index examiner s advice throughout so you will know what to expect in the exam and will be able to demonstrate the skills required exam style questions with graded student responses so you can see clearly what is required to get a better grade OCR(A) A2 Chemistry Student Unit Guide: Unit F325 Equilibria, Energetics and Elements Mike Smith, 2009-06-26 Student Unit Guides are perfect for revision Each guide is written by an examiner and explains the unit requirements summarises the relevant unit content and includes a series of specimen questions and answers There are three sections to each guide Introduction includes advice on how to use the guide an explanation of the skills being tested by the assessment objectives an outline of the unit or module and depending on the unit suggestions for how to revise effectively and prepare for the examination questions Content Guidance provides an examiner s overview of the module s key terms and concepts and identifies opportunities to exhibit the skills required by the unit It is designed to help students to structure their revision and make them aware of the concepts they need to understand the exam and how they might analyse and evaluate topics Question and Answers sample questions and with graded answers which have been carefully written to reflect the style of the unit All responses are accompanied by commentaries which highlight their respective strengths and weaknesses giving students an insight into the mind of the examiner Revise As and A2 - Chemistry Rob Ritchie, 2008-10 *Revise AS A2 Chemistry* gives complete study support throughout the two A Level years This Study Guide matches the curriculum content and provides in depth course coverage plus invaluable advice on how to get the best results in the exams *Valuing Labour in Greco-Roman Antiquity* , 2024-03-11 How did ancient Greeks and Romans regard work It has long been assumed that elite thinkers disparaged physical work and that working people rarely commented on their own labors The papers in this volume challenge these notions by investigating philosophical literary and working people s own ideas about what it meant to work From Plato s terminology of labor to Roman prostitutes self proclaimed pride in their work these chapters find ancient people assigning value to multiple different kinds of work and many different concepts of labor *Cumulated Index Medicus* , 1998 **Commerce Business Daily** , 2001-05-22 **New Zealand National Bibliography to the Year 1960: 1889-1960, A-H** Austin Graham Bagnall, 1970 *New Zealand National Bibliography* , 1969 **New Zealand National Bibliography to the Year 1960** Austin Graham Bagnall, 1970 Dictionary Catalog Schomburg Collection of Negro Literature and History, 1962

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