

SNC2DI - REVIEW FOR CHEMISTRY

- Review "Key Terms" from Chapters 4, 5, & 6 (p.139, 193, 221)
- Review "Unit Summary" listing key concepts, chapter summaries, vocabulary, and key visuals on pg 244
- Try "Chapter Review" questions:
 - Ch. 4 → pg 190-191 #4-6, 8, 11, 12-17, 19-20
 - Ch. 5 → pg 218-219 #1-3, 9-11, 13
 - Ch. 6 → pg 242-243 #1, 3, 6-8, 10-12, 14-20
- The following "Unit Review" questions:
 - pg 248-249 #2-11, 18-20, 25-27, #30-32
 - pg 250-251 #43-44, 49-54, 59
 - pg 252-253 #60-61
- Study your notes/quizzes/labs from the beginning of the unit. Getting them organized first will make this much easier to do!!
- Make study notes to organize the information.

TOPIC LIST

- Periodic table
- Bohr diagrams of atoms & ions
- Ions and the periodic table
- Chemical families names on the periodic table and their charges
- Ionic compounds naming & formulas
- Polyatomic compounds naming & formulas
- Molecular compounds naming and formulas
- Types of chemical reactions (five)
- Law of conservation of mass
- Counting atoms
- Balancing equations
- Acids & bases
- pH scale

If you have any questions, please ask! I am here to help. Please come in at lunch or set up an appointment to meet before or after school if you would like additional help outside class time.

Good luck and happy studying!

Chemistry Unit Review: ANSWERS

Unit Review Questions from Text:

p.330-332 # 1a-f, 3 a-g, 4-7, 10, 11, 12, 13, 14, 15, 16, 18

Atoms

1. Fill in the blanks: "Nucleus with positive protons and neutral neutrons. Negative electrons in orbits outside nucleus. The electrons in the outermost orbit are called valence electrons, and they are involved in chemical reactions. If electrons are lost, then a positive ion forms, and if electrons are gained, then a negative ion forms."

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