

**SECTION A****QUESTION 1**

**1.1** Various possible options are provided as answers to the following questions. Choose the correct answer and write only the letter (A – D) next to the question (1.1.1 – 1.1.5). For example 1.1.5 D.

1.1.1 The basic functional unit of the environment is the...

- A. biosphere.
- B. ecosystem.
- C. population.
- D. lithosphere.

1.1.2 All organisms around the world that live in the same climate share the same...

- A. population.
- B. ecosystem.
- C. biome.
- D. community.

1.1.3 Which of the following processes occur during the nitrogen cycle?

- (i) Consumption of plant protein by herbivores.
- (ii) The decay of dead organisms by decomposers.
- (iii) The conversion of nitrates to nitrites by bacteria.
- (iv) The absorption of nitrates by plants.

- A. (i), (ii) and (iii)
- B. (ii), (iii) and (iv)
- C. (i), and (iv)
- D. (i), (ii) and (iv)

1.1.4 When a jackal kills and eats a rabbit, the jackal is the...

- A. Producer.
- B. Prey.
- C. Predator.
- D. Saprophyte.

1.1.5 Which one of the following is not an abiotic factor?

- A. Height above sea level.
- B. Oxygen in the atmosphere.
- C. Water vapour in the atmosphere.
- D. Bacteria in the soil.

(2 x 5) (10)

# Maths Paper 1 Grade 11 Caps Exemplar 2014

**Albert A Gayle**



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web answer  $\text{Na}^+ \text{aq}$  and  $\text{SO}_4^{2-} \text{aq}$  identify the chemical equation the spectator ions  $\text{KOH} \text{aq}$   $\text{HNO}_3$   $\text{KNO}_3 \text{aq}$   $\text{H}_2\text{O}$   $\text{K}^+ \text{aq}$   $\text{OH}^- \text{aq}$   $\text{H}^+ \text{aq}$   $\text{NO}_3^- \text{aq}$   $\text{K}^+ \text{aq}$   $\text{NO}_3^- \text{aq}$   $\text{H}_2\text{O}$  l answer  $\text{K}^+ \text{aq}$  and  $\text{NO}_3^- \text{aq}$  identify the chemical equation the spectator ions  $\text{Pb(NO}_3)_2 \text{aq}$   $2\text{KI} \text{aq}$   $\text{PbI}_2 \text{s}$   $2\text{KNO}_3 \text{aq}$

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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction  $2\text{Ag}^+ \text{aq}$   $\text{Cr}_2\text{O}_7^{2-} \text{aq}$



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web jan 29 2020 the ionic form of this reaction is  $2 \text{Na}^+ \text{aq} + 2 \text{Cl}^- \text{aq} + \text{Cu}^{2+} \text{aq} + \text{SO}_4^{2-} \text{aq}$  the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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web the six  $\text{Na}^+$  ions and the six  $\text{Cl}^-$  ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation  $3\text{Ba}^{2+} \text{aq} + 2\text{PO}_4^{3-} \text{aq} \rightarrow \text{Ba}_3(\text{PO}_4)_2 \text{s}$

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water  $\text{H}^+ \text{aq} + \text{OH}^- \text{aq} \rightarrow \text{H}_2\text{O} \text{l}$

2 of the spectator ions in  $\text{KOH(aq)}$  and  $\text{HNO}_3\text{(aq)}$  are  $\text{K}^+\text{(aq)}$  and  $\text{NO}_3^-\text{(aq)}$  how to identify and list the

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$2\text{Cl}^-\text{aq} + \text{Cu}^{2+}\text{aq} + 2\text{Cl}^-\text{aq} + \text{Cu}^{2+}\text{aq}$  these spectator ions are not actively involved in the reaction the existence of spectator ions affects the Debye length

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