

Name: _____ Date: _____ Per: _____

Ionic Bonding WebQuest

Part 1: Go to http://www.visionlearning.com/library/module_viewer.php?mid=55

Read the **Chemical Bonding** and **Ionic Bonding** sections, and answer the following questions:

1. Elements react with each other to form _____ and _____.
2. What two elements is table salt composed of? _____ and _____.
3. Who contributed to our knowledge of valence electrons? _____.
4. What are the two types of bonding? _____ and _____.
5. Briefly describe how an ionic bond forms: _____.

6. Look at the picture of the **NaCl Crystal Schematic**. Explain why there is no such thing as a single molecule of salt.

7. Click the link [The reaction of sodium with chlorine](#) and watch through the video there. Briefly describe each of the steps involved with the reaction of sodium with chlorine.

Part 2: Watch the following clips on Ionic Bonding:

- <http://search2video.net/ionic-bonding-video-ofRtGtkaGwKP.html>
- <http://www.youtube.com/watch?v=QajcDyz8W0s> (ignore the covalent part)
- http://www.mhhe.com/physac/chemistry/animations/chem_7a_ssp/comp1a2_11.swf (you can click on the NaF ionic compound and see an animation on how it forms.)

Part 3: Explore the link below. (You can click on the elements and compounds in the video for extra info.)
http://content.tutorvista.com/chemistry_10/content/media/048_electro_bonding.swf

8. Describe how ionic compounds form.

Part 4: Take the quiz at the following website. Record your score here: _____ of 10.
http://www.softschools.com/quizzes/chemistry/ionic_bonding_formation/quiz993.html

Part 5: Dissociation of Salt (an ionic compound). Watch the following videos:

- <http://www.youtube.com/watch?v=EBtGcTAuF4o>
- <http://www.mhhe.com/physac/chemistry/essentialchemistry/flash/molvie1.swf>

9. Go to <http://phet.colorado.edu/en/simulation/sugar-and-salt-solutions>. Use the simulation and explain how salt and sugar dissolve in water.

Ionic Covalent Bonding Webquest Answer Key

Rachel Turoscy



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Table of Contents Ionic Covalent Bonding Webquest Answer Key

1. Understanding the eBook Ionic Covalent Bonding Webquest Answer Key
 - The Rise of Digital Reading Ionic Covalent Bonding Webquest Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Ionic Covalent Bonding Webquest Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ionic Covalent Bonding Webquest Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ionic Covalent Bonding Webquest Answer Key
 - Personalized Recommendations
 - Ionic Covalent Bonding Webquest Answer Key User Reviews and Ratings
 - Ionic Covalent Bonding Webquest Answer Key and Bestseller Lists
5. Accessing Ionic Covalent Bonding Webquest Answer Key Free and Paid eBooks
 - Ionic Covalent Bonding Webquest Answer Key Public Domain eBooks
 - Ionic Covalent Bonding Webquest Answer Key eBook Subscription Services
 - Ionic Covalent Bonding Webquest Answer Key Budget-Friendly Options

6. Navigating Ionic Covalent Bonding Webquest Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Ionic Covalent Bonding Webquest Answer Key Compatibility with Devices
 - Ionic Covalent Bonding Webquest Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ionic Covalent Bonding Webquest Answer Key
 - Highlighting and Note-Taking Ionic Covalent Bonding Webquest Answer Key
 - Interactive Elements Ionic Covalent Bonding Webquest Answer Key
8. Staying Engaged with Ionic Covalent Bonding Webquest Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ionic Covalent Bonding Webquest Answer Key
9. Balancing eBooks and Physical Books Ionic Covalent Bonding Webquest Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ionic Covalent Bonding Webquest Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ionic Covalent Bonding Webquest Answer Key
 - Setting Reading Goals Ionic Covalent Bonding Webquest Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ionic Covalent Bonding Webquest Answer Key
 - Fact-Checking eBook Content of Ionic Covalent Bonding Webquest Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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