

Name Solutions

Date _____ Period _____

Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
Li	1 ve	transfers $1e^-$ to F	Li^+
F	7 ve	accepts $1e^-$ from Li	F^-

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
1) $Li + Cl \Rightarrow LiCl$ $Li + Cl \rightarrow [Li]^+ [Cl]^-$	Li	1 ve	donate $1e^-$	Li^+
	Cl	1 ve	accepts $1e^-$	Cl^-
2) $Ca + O \Rightarrow CaO$ $Ca + O \rightarrow [Ca]^{+2} [O]^{2-}$	Ca	2 ve	donates $2e^-$	Ca^{+2}
	O	6 ve	accepts $2e^-$	O^{2-}
3) $Be + F \Rightarrow BeF_2$ $Be + 2F \rightarrow [Be]^{+2} [F]_2^{2-}$	Be	2 ve	donates $2e^-$ to each F	Be^{+2}
	F	7 ve	each F accepts $1e^-$ each	F^-
4) $Mg + S \Rightarrow MgS$ $Mg + S \rightarrow [Mg]^{+2} [S]^{2-}$	Mg	2 ve	donates $2e^-$ to S	Mg^{+2}
	S	6 ve	accepts $2e^-$	S^{2-}
5) $K + F \Rightarrow KF$ $K + F \rightarrow [K]^+ [F]^-$	K	1 ve	donates $1e^-$	K^+
	F	7 ve	accepts $1e^-$	F^-

Ionic Bonding Worksheet Answer Key

Daniela Niemeyer

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Ionic Bonding Worksheet Answer Key:

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Table of Contents Ionic Bonding Worksheet Answer Key

1. Understanding the eBook Ionic Bonding Worksheet Answer Key
 - The Rise of Digital Reading Ionic Bonding Worksheet Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Ionic Bonding Worksheet Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ionic Bonding Worksheet Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ionic Bonding Worksheet Answer Key
 - Personalized Recommendations
 - Ionic Bonding Worksheet Answer Key User Reviews and Ratings
 - Ionic Bonding Worksheet Answer Key and Bestseller Lists
5. Accessing Ionic Bonding Worksheet Answer Key Free and Paid eBooks
 - Ionic Bonding Worksheet Answer Key Public Domain eBooks
 - Ionic Bonding Worksheet Answer Key eBook Subscription Services
 - Ionic Bonding Worksheet Answer Key Budget-Friendly Options

6. Navigating Ionic Bonding Worksheet Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Ionic Bonding Worksheet Answer Key Compatibility with Devices
 - Ionic Bonding Worksheet Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ionic Bonding Worksheet Answer Key
 - Highlighting and Note-Taking Ionic Bonding Worksheet Answer Key
 - Interactive Elements Ionic Bonding Worksheet Answer Key
8. Staying Engaged with Ionic Bonding Worksheet Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ionic Bonding Worksheet Answer Key
9. Balancing eBooks and Physical Books Ionic Bonding Worksheet Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ionic Bonding Worksheet Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ionic Bonding Worksheet Answer Key
 - Setting Reading Goals Ionic Bonding Worksheet Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ionic Bonding Worksheet Answer Key
 - Fact-Checking eBook Content of Ionic Bonding Worksheet Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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