STO.1	Balance a chemical equation.	
STO.2	Identify the parts of a chemical equation.	
BXN.1	Describe a chemical reaction using words and symbolic equations.	

For each of the following problems, write complete chemical equations to describe the chemical process taking place. Balance the equations.

- When lithium hydroxide pellets are added to a solution of sulfuric acid (H<sub>2</sub>SO<sub>4</sub>), Ithium sulfate and water are formed.
- Magnesium reacts with sodium fluoride to produce magnesium fluoride and elemental sodium.
- If a copper coil is placed into a solution of silver nitrate, silver crystals form and copper (I) nitrate is generated.
- When crystalline C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> is burned in oxygen, carbon dioxide and water vapor are formed.
- Calcium carbonate combines with hydrochloric acid (HCI) to produce calcium chloride, water and carbon dioxide gas.

Balance the equations below:

8) 
$$C_3H_8 + C_2 \rightarrow CO_2 + H_2O$$

11) 
$$P + Q_2 \rightarrow P_2O_5$$

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Shasha Hu

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