Macromolecules are individual large polymer molecules.

A functional group is a chemically reactive atom or group of atoms that imparts characteristic properties to organic compounds containing that group.

For example: The carbonyl group C = O is the functional group for ketones.

Free radicals are very reactive atoms or molecules which have unpaired electrons.

The bond between the two coygen atoms breaks first as it is not as stable as the bond between the O-H atoms,

Polylactic acid PLA

H

0

Condensation (esterification) of lactic acid to form lactide.

O

Direct or self-polymerization is not a good option as too many water molecules are liberated,

Lactic acid is 2-hydroxypropanoic acid

Lactose is the predominant sugar in milk and enzymes turn it into lactic acid.

When a chain of lactic acid undergoes self-polymerisation, too much water is produced by the esterification (elimination) reactions.

0

It is far more efficient to place two monomers next to each other with one upside down so that the carboxylic and hydroxy functional groups are next to each other. An elimination reaction occurs as water is eliminated to produce lactide.

No initiator is needed. The reaction is slower because two steps are needed. Lactide must first form and then be polymerised. No double bonds are needed as the acid and alcohol functional groups react by eliminating water.

Holt Chemistry Carbon And Organic Compounds Answers

Clemens Wendtner

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Table of Contents Holt Chemistry Carbon And Organic Compounds Answers

- 1. Understanding the eBook Holt Chemistry Carbon And Organic Compounds Answers
 - The Rise of Digital Reading Holt Chemistry Carbon And Organic Compounds Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Holt Chemistry Carbon And Organic Compounds Answers
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Chemistry Carbon And Organic Compounds Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Holt Chemistry Carbon And Organic Compounds Answers
 - Personalized Recommendations
 - Holt Chemistry Carbon And Organic Compounds Answers User Reviews and Ratings

- Holt Chemistry Carbon And Organic Compounds Answers and Bestseller Lists
- 5. Accessing Holt Chemistry Carbon And Organic Compounds Answers Free and Paid eBooks
 - Holt Chemistry Carbon And Organic Compounds Answers Public Domain eBooks
 - Holt Chemistry Carbon And Organic Compounds Answers eBook Subscription Services
 - Holt Chemistry Carbon And Organic Compounds Answers Budget-Friendly Options
- 6. Navigating Holt Chemistry Carbon And Organic Compounds Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Holt Chemistry Carbon And Organic Compounds Answers Compatibility with Devices
 - Holt Chemistry Carbon And Organic Compounds Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Chemistry Carbon And Organic Compounds Answers
 - Highlighting and Note-Taking Holt Chemistry Carbon And Organic Compounds Answers
 - Interactive Elements Holt Chemistry Carbon And Organic Compounds Answers
- 8. Staying Engaged with Holt Chemistry Carbon And Organic Compounds Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Chemistry Carbon And Organic Compounds Answers
- 9. Balancing eBooks and Physical Books Holt Chemistry Carbon And Organic Compounds Answers
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Holt Chemistry Carbon And Organic Compounds Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Holt Chemistry Carbon And Organic Compounds Answers
 - Setting Reading Goals Holt Chemistry Carbon And Organic Compounds Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Holt Chemistry Carbon And Organic Compounds Answers
 - Fact-Checking eBook Content of Holt Chemistry Carbon And Organic Compounds Answers
 - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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