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Covalent Bonding

To gain a stable electron arrangement, non metal elements share their outer electrons. This is called _____ bonding.

To be stable, an atom needs to have **8** electrons in its outer electron shell. Neon is a noble gas.



The noble gases are the only elements that are _____ on their own.

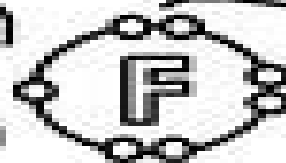
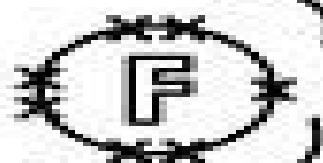
non-metals

are on the _____ hand side of the periodic table.

these are called "dot and cross" diagrams

each **X** and each **o** represents an electron

example



the positive nucleus of each atom attracts its own electrons and the electrons from the other atom

the Fluorine atoms are unstable on their own

when they share electrons, each atom has 8 outer electrons = making them stable



covalent substances can be

covalent molecular
or
covalent network

Glencoe Chemistry Guided Notes Chapter 9 Covalent Bonding

Rachel Sandford



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