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AW Chickering



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4 notify your therapist if you have any questions or concerns 5

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web key points a brønsted lowry acid is any species that is capable of donating a proton h a brønsted lowry base is any species that is capable of accepting a proton which requires a lone pair of electrons to bond to the h water is amphoteric which means it can act as both a brønsted lowry acid and a brønsted lowry base

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we think of them in pairs called conjugate pairs when the acid H_3O^+ loses a proton it forms a base H_2O when the base H_2O accepts a proton back again it obviously reforms the acid H_3O^+

web the key to understanding acids as well as bases and salts had to await michael faraday s mid nineteenth century discovery by 1890 the swedish chemist svante arrhenius 1859 1927 was able to formulate the first useful theory of acids arrhenius definition an acidic substance is one whose molecular unit contains at least one

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