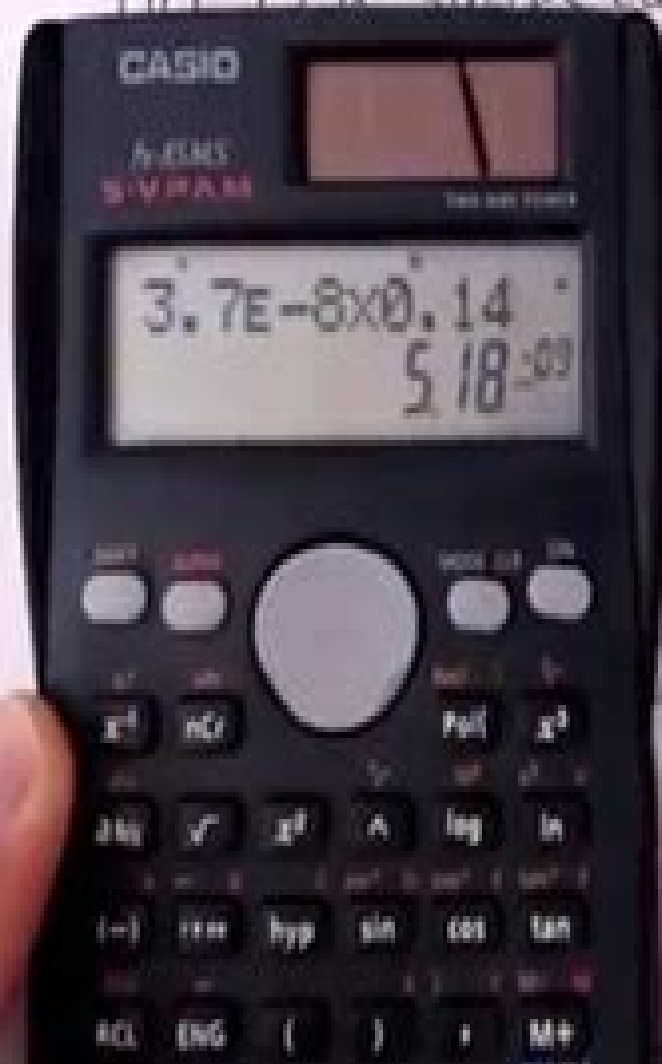


- (a) 0.14 mol dm^{-3} solutions of hydrochloric acid, HCl , and chloric(III) acid, HClO_3 (a, 1.74), have different pH values.

Explain why the pH values are different and calculate the pH of 0.14 mol dm^{-3} solutions of HCl and HClO_3 to two decimal places.

Show any working in calculations.

HCl is a strong acid and HClO_3 is



$$\rightarrow -\log 0.14 = 0.85.$$

$$7 \times 10^{-8}$$

$$[\text{H}^+]^2 = 3.7 \times 10^{-8} \times 0.14$$

$$= 5.18 \times 10^{-9}$$

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David Baud



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5.5 - Hypothesis Testing for Two-Sample Proportions

We are now going to develop the hypothesis test for the difference of two proportions for independent samples. The hypothesis test follows the same steps as ...

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1.3.5.3. Two-Sample t-Test for Equal Means Purpose: Test if two population means are equal

The two-sample t-test (Snedecor and Cochran, 1989) is used to determine if two population means are equal.

2 Sample t-Test (1 tailed)

Suppose we have two samples of ceramic sherd thickness collected from an archaeological site, where the two samples are easily distinguishable by the use of.

Two sample t-test: SAS instruction

Note that the test is two-sided (sides=2), the significance level is 0.05, and the test is to compare the difference between two means ($\mu_1 - \mu_2$) against 0 (H_0 ...

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