

- (v) What is the action of selenium on magnesium metal?
- (vi) Write the name of isomerism in the following complexes :
 $[\text{Cu}(\text{NH}_3)_4]$ $[\text{PtCl}_4]$ and $[\text{Pt}(\text{NH}_3)_4][\text{CuCl}_4]$
- (vii) Write the name of the alloy used in Fischer Tropsch process in the synthesis of gasoline.
- (viii) Henry's law constant for $\text{CH}_3\text{Br}_{(\text{g})}$ is $0.159 \text{ mol dm}^{-3} \text{ bar}^{-1}$ at 25°C . What is solubility of $\text{CH}_3\text{Br}_{(\text{g})}$ in water at same temperature and partial pressure of 0.164 bar ?

SECTION - B

Attempt any **EIGHT** of the following questions :

[16]

- Q. 3. Explain pseudo-first order reaction with suitable example.
- Q. 4. Write the consequences of Schottky defect with reasons.
- Q. 5. What is the action of following on ethyl bromide :
 (i) Na in dry ether
 (ii) Mg in dry ether
- Q. 6. Explain formation of peptide linkage in protein with an example.
- Q. 7. Derive an expression to calculate molar mass of non volatile solute by osmotic pressure measurement.
- Q. 8. Explain monodentate and ambidentate ligands with example.
- Q. 9. Explain the trends in the following atomic properties of group 16 elements :
 (i) Atomic radii
 (ii) Ionisation enthalpy
 (iii) Electronegativity
 (iv) Electron gain enthalpy

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