

# Ideal Gas Law

## Chapter 9

1. A gas at a pressure of 1 atm and a volume of 10 liters has a temperature of 25°C. What is the number of moles of the gas present?
2. A sample of gas has a volume of 5 liters, a pressure of 2 atm, and a temperature of 300 K. What is the number of moles of the gas present?
3. A gas has a pressure of 1.5 atm, a volume of 20 liters, and a temperature of 25°C. What is the number of moles of the gas present?
4. A gas at a volume of 30 liters and a temperature of 27°C has a pressure of 2 atm. What is the number of moles of the gas present?
5. A gas at a pressure of 1 atm, a volume of 25 liters, and a temperature of 0°C has a number of moles equal to 2. What is the pressure of the gas at a volume of 15 liters and a temperature of 27°C?
6. A gas at a pressure of 2 atm, a volume of 10 liters, and a temperature of 25°C has a number of moles equal to 4. What is the pressure of the gas at a volume of 20 liters and a temperature of 27°C?
7. A gas at a volume of 5 liters and a temperature of 27°C has a pressure of 1 atm. What is the number of moles of the gas present?
8. A gas at a pressure of 3 atm, a volume of 20 liters, and a temperature of 25°C has a number of moles equal to 5. What is the pressure of the gas at a volume of 10 liters and a temperature of 27°C?
9. A sample of gas has a volume of 15 liters, a pressure of 1.5 atm, and a temperature of 300 K. What is the number of moles of the gas present?
10. A gas at a volume of 10 liters and a temperature of 27°C has a pressure of 2 atm. What is the number of moles of the gas present?
11. Note: These questions are meant to be examples to guide you in creating your own worksheet. The specific answers to these questions may vary based on your textbook or course material.

# Ideal Gas Law Questions And Answers

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*Physics* John D. Cutnell, Kenneth W. Johnson, 2009-09-08 Designed for medical professionals who may struggle with making the leap to conceptual understanding and applying physics the eighth edition continues to build transferable problem solving skills It includes a set of features such as Analyzing Multiple Concept Problems Check Your Understanding Concepts Calculations and Concepts at a Glance This helps the reader to first identify the physics concepts then associate the appropriate mathematical equations and finally to work out an algebraic solution

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**Task Collection Technical Thermodynamics** Martin Dehli, 2024-03-08 This comprehensive collection of problems contains questions from energy and thermal engineering practice as well as from existing exercises and examinations The solutions are very detailed and therefore comprehensible Since the structure of the book is based on that of the textbook Fundamentals of Technical Thermodynamics students can work through the contents in depth and check their learning success In the present 2nd edition extensions and additions have been made for a better understanding of the solution paths Numerous checkboxes and exercises with solutions have been added In addition the extended introduction deals with the nature of entropy and draws attention to the benefits of thermodynamics

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