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Hunter Automotive Lift Manual

Emilie Sanchez



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web this is like a one dimensional system whose mechanical energy e is a constant and whose potential energy with respect to zero energy at zero displacement from the spring's unstretched length $x = 0$ is $U = \frac{1}{2} kx^2$ figure 8 5 2 8 5 2 a a glider between springs on an air track is an example of a horizontal mass spring system

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web 1 if the pe of the reactants is 430kj and the pe of the products is 190kj what is the Δh 2 if the pe of the reactants is 430kj and the activation energy is 220 how much total energy will be needed to be added for the reaction to proceed transcribed image text potential energy diagram answer the questions using the graph

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web figure 5 5 1 a potential energy diagram shows the total potential energy of a reacting system as the reaction proceeds a in an endothermic reaction the energy of the products is greater than the energy of the reactants and Δh is positive b in an exothermic reaction the energy of the products is lower than the energy of the

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constant of gases t is the absolute temperature

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into products figure below shows basic potential energy diagrams for an endothermic a and an exothermic b reaction recall that the enthalpy change Δh is positive for an endothermic reaction and negative for an

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web where the salt is the conjugate base of the acid the four parts of the titration curve are described below and you should look to the appropriate text section to see how they are treated pure acid 0 mL of base is added section 17.3.2.1

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Volume of NaOH added (mL)	pH of the solution
4	6.4
6	4.6
4	6.5
9	5.9
5	9.5
9	4.6

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