

Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 1

Q1. At the same temperature and pressure, an equal volume of gases contains the same number of ____.

- a.) Molecules
- b.) Electrons
- c.) Protons
- d.) Particles

Q2. The gas law that relates the volume of a gas to the number of molecules of the gas is ____.

- a.) Gay-Lussac's law
- b.) Avogadro's law
- c.) Boyle's law
- d.) Charles's law

Q3. The mass % of carbon in CO_2 is ____.

- a.) 0.034%
- b.) 27.27%
- c.) 3.4%
- d.) 28.7%

Q4. The empirical formula of benzene is ____.

- a.) CH
- b.) C_6H_6
- c.) C_3H_3
- d.) C_2H_2

Q5. What is the mole fraction of the solute in a 1.00 molal aqueous solution?

- a.) 0.00177
- b.) 0.0344
- c.) 0.0177
- d.) 0.1770

Q6. State the number of significant figures in each of the following numbers:

- i.) 207.35

Hrw Chemistry Ch 1

**René Lahaie, Luc Papillon, Pierre
Valiquette**



Hrw Chemistry Ch 1:

Unveiling the Energy of Verbal Artistry: An Mental Sojourn through **Hrw Chemistry Ch 1**

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