

## Quiz 4 – Potential Energy Diagrams

1. The addition of a catalyst to a reaction provides an alternative mechanism with

- A. Lower activation energy and lower reaction rate
- B. Lower activation energy and higher reaction rate**
- C. Higher activation energy and lower reaction rate
- D. Higher activation energy and higher reaction rate

2. Consider the following Reaction



The correct equation including the heat term is

- A.  $\text{N}_{2(g)} + \text{O}_{2(g)} + 90 \text{ kJ} \longrightarrow 2\text{NO}_{(g)}$
- B.  $\text{N}_{2(g)} + \text{O}_{2(g)} + 180 \text{ kJ} \longrightarrow 2\text{NO}_{(g)}$**
- C.  $\text{N}_{2(g)} + \text{O}_{2(g)} \longrightarrow 2\text{NO}_{(g)} + 90 \text{ kJ}$
- D.  $\text{N}_{2(g)} + \text{O}_{2(g)} \longrightarrow 2\text{NO}_{(g)} + 180 \text{ kJ}$

3. A forward reaction has an activation energy of 50 kJ and a  $\Delta H$  of  $-100 \text{ kJ}$ .  
The PE diagram, which describes this reaction, is

A.

4. Consider the following potential energy diagram

The Activation energy for the forward reaction is

- A. 25 kJ
- B. 50 kJ
- C. 75 kJ**
- D. 125 kJ

5. Consider the following reaction



# Multiple Choice Questions Energy Diagrams

**L Reisser**



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Services, 2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter's end Two full length practice tests with detailed answer explanations are included in the book **Problems and Problem Solving in Chemistry Education**

Georgios Tsaparis, 2021-05-17 Problem solving is central to the teaching and learning of chemistry at secondary tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner

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## **Table of Contents Multiple Choice Questions Energy Diagrams**

1. Understanding the eBook Multiple Choice Questions Energy Diagrams
  - The Rise of Digital Reading Multiple Choice Questions Energy Diagrams
  - Advantages of eBooks Over Traditional Books
2. Identifying Multiple Choice Questions Energy Diagrams
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Multiple Choice Questions Energy Diagrams
  - User-Friendly Interface
4. Exploring eBook Recommendations from Multiple Choice Questions Energy Diagrams
  - Personalized Recommendations
  - Multiple Choice Questions Energy Diagrams User Reviews and Ratings
  - Multiple Choice Questions Energy Diagrams and Bestseller Lists

5. Accessing Multiple Choice Questions Energy Diagrams Free and Paid eBooks
  - Multiple Choice Questions Energy Diagrams Public Domain eBooks
  - Multiple Choice Questions Energy Diagrams eBook Subscription Services
  - Multiple Choice Questions Energy Diagrams Budget-Friendly Options
6. Navigating Multiple Choice Questions Energy Diagrams eBook Formats
  - ePub, PDF, MOBI, and More
  - Multiple Choice Questions Energy Diagrams Compatibility with Devices
  - Multiple Choice Questions Energy Diagrams Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Multiple Choice Questions Energy Diagrams
  - Highlighting and Note-Taking Multiple Choice Questions Energy Diagrams
  - Interactive Elements Multiple Choice Questions Energy Diagrams
8. Staying Engaged with Multiple Choice Questions Energy Diagrams
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Multiple Choice Questions Energy Diagrams
9. Balancing eBooks and Physical Books Multiple Choice Questions Energy Diagrams
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Multiple Choice Questions Energy Diagrams
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Multiple Choice Questions Energy Diagrams
  - Setting Reading Goals Multiple Choice Questions Energy Diagrams
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Multiple Choice Questions Energy Diagrams
  - Fact-Checking eBook Content of Multiple Choice Questions Energy Diagrams
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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