Unit 1 Measurements

Highest Mastery Achieved for 2.2.2

(80% considered mastered)

Things to know: scientific method, qualitative/quantitative data, Si measurements (m, L, g), ways to findvolume, converting with prefixes, density, accuracy vs. precision, % error, significant figures, scientific notation, atom/element/compound, chemical/physical properties, chemical/physical change, classification of matter (mixture/pure substance and homogeneous/heterogeneous), periodic table (periods/groups/families, metals/nonmetals/metalloids)

*Items in Italics is for honors only

Objective

2.2.2—Analyzethe evidence of chemical change.

Problems

- List the four indicators of a chemical change:
 - a. Formation of a precipitate
 - b. Formation of a gas
 - c. Change in color
 - d. Change in energy
- Define precipitate. How can you determine if something is a precipitate or not?
 New solid formed during a reaction...is always on the product side (right side)...can

determine a precipitate by using the solubility rules (insoluble = precipitate)

- What would happen if a burning splint was exposed to:
 - Oxygen the flame would be brighter and bigger
 - b. Hydrogen the flamewould make a popping sound...sometimes called barking
 - c. Carbon dioxidethe flame would go out
- 4. What would happen if carbon dioxide was bubbled through lime water?

The water would look cloudy

5. What piece of safety equipment must be used in all laboratory experiments?

Goggles.

6. When using a Bunsen burner what precautions should be made?

Long hair and clothes should be pulled back and there should be no flammable materials near by

- Exothermic reaction releases energy, AH is negand energy is shown on the product (right) side.
- Endothermic reaction absorbesenergy, ΔH is +and energy is shown on the reactant (left)side.
- 9. Label each of the following as a physical or chemical property and give a brief reason
 - a. Flammability c. Can neutralize a base CP, cannot get substance back/change in color CP, cannot get substance back
 - b. Density
 PP, description_doesn't change substance
 PP, phase change_still same substance
- 10. Label each of the following as a physical or chemical change
 - a. A tire is inflated with air.
 c. Water is added to red soln, it turns pink.

PC, doesn't change substance PP, diluted_same substance_no color change

- Food is digested in the stomach.
 d. Water is heated and changed into steam.
 - CP, can't get substance back, gas formed PP, phase change...still same substance

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Tobias Bleicker



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