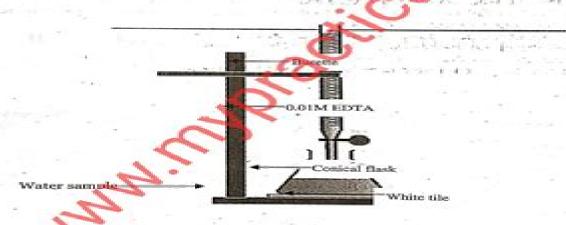
alkaline earth metal ions Ca<sup>2+</sup> and Mg<sup>2+</sup>. The EDTA reagent can be used to measure the total quantity of dissolved Ca<sup>2+</sup> and Mg<sup>2+</sup> ions in a water sample.

Determination of hardness of water is based on the fact that when indicator.

Determination of hardness of water is based on the fact that when indicator, Eriochrome Black -T (EBT) is added to hard water in alkaline medium, it forms wine red colour complex with Ca<sup>2\*</sup> and Mg<sup>2\*</sup> ions

Thus the total hardness of a water sample can be estimated by titration with standard solution of EDTA.

#### VIII Practical set-up / Circuit diagram / Work Situation



#### IX Resources Required

S. No.	Name of Resource	Suggested Broad Specification	
1.	Burette	Borosil glass . Capacity 25 mt/se	Quantity
2.	Conical flask	Borosil plass. Canacire 250 11100	11. Prof. B. 1965. 12. 1962. 11. 1975. 11. 11. 11. 11. 11. 11. 11. 11. 11. 1
3.	Beaker	Borosil glass Capacity 250	
4.	Pipette	Borosil glass, Capacity 25-14	
5.	EDTA solution	0.01 M	
6.	Buffer solution	pH = 10	As per Requirement
7.	Eriochrome Black T indicator		
8.	Water sample	Hard water sample	

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