

# *Joomla Beginners Guide*



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# Joomla 2beginner39s Guide

**Mark Dexter,Louis Landry**



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web we utilize the equation that relate amount remaining initial mass and number of half lives  $n = \frac{t}{t_{1/2}} \times \log_2 \frac{N_0}{N}$

125 g 5 what is the half life of an isotope that is 80 remained after 16 days solution remaining 80 100 therefore  $n = \frac{t}{t_{1/2}} \times \log_2 \frac{N_0}{N}$

now using the half life equation in example 1 we have

web what is the half life of this radioisotope after 2 days 5 grams of the radioisotope 1 day b 2 days c 5 days d 4 days 3 the ratio of uranium 238 to lead 206 in a mineral is used to determine age b density c solubility d composition 4 a radioactive isotope has a half life of 10 years

web chapter 15 nuclear chemistry half life learning objectives define half life determine the amount of radioactive substance remaining after a given number of half lives whether or not a given isotope is radioactive is a characteristic of that particular isotope

web the half life of a radioactive isotope is the amount of time it takes for one half of the radioactive isotope to decay the half life of a specific radioactive isotope is constant it is unaffected by conditions and is independent of the initial amount of that isotope

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terms half life and the rate of radioactive decay unstable nuclei decay

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