# CLASSIFYING CHEMICAL REACTIONS



Classify the following reactions as synthesis, decomposition, single replacement or double replacement. De composition 2KCIO, → 2KCI + 3O, Double Replacement HCI + NaOH → NaCI + H<sub>\*</sub>O Single Replacement. Mg + 2HCl → MgCl, + H, Synthesis 2H<sub>2</sub> + O<sub>2</sub> → 2H<sub>2</sub>O Single Replacement 2AI + 3NIBr<sub>2</sub> → 2AIBr<sub>3</sub> + 3NI Synthesis 4AI + 3O<sub>5</sub> → 2AI<sub>2</sub>O<sub>5</sub> Decomposition 2NaCl → 2Na + Cl<sub>a</sub> Single Replacement CaCl<sub>2</sub> + F<sub>2</sub> → CaF<sub>2</sub> + Cl<sub>2</sub> Double Replacement AgNO<sub>3</sub> + KCI → AgCI + KNO<sub>3</sub> Synthesis N<sub>2</sub> + 3H<sub>2</sub> → 2NH<sub>3</sub> Decomposition 2H<sub>2</sub>O<sub>3</sub> → 2H<sub>2</sub>O + O<sub>3</sub> 12. (NH,)250, + Ba(NO3)2 - Baso, + 2NH,NO, Double Replacement Single Replacement 13. Mgl₂ + Br₂ → MgBr₂ + l₂ Synthesis 14. SO, + H₂O → H₂SO₄

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# **Identifying Chemical Reactions Answer Key**

**AW Rasmussen** 

**Identifying Chemical Reactions Answer Key:** 

Unveiling the Magic of Words: A Review of "Identifying Chemical Reactions Answer Key"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "Identifying Chemical Reactions Answer Key," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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