

HSC Maharashtra Board question paper: March 2013

Note:

- All questions are compulsory
- Answer to the two sections are to be written in the same answer book.
- Figure to the right hand side indicate full marks.
- Write balanced chemical equations and draw neat and labelled diagrams wherever necessary.
- Every new question must be started on a new page.
- Use of logarithmic table is allowed.

CHEMISTRY: SECTION – II

Q. 5. Select and write the most appropriate answer from the given alternatives for each sub-question: [7]

- In which pair highest oxidation states of transition metals are found:
 (A) nitriles and chlorides (B) fluorides and chlorides
 (C) fluorides and oxides (D) nitriles and oxides

- Which of the following carbocations is least stable?

- (A) $\text{CH}_3 - \text{CH}_2 - \overset{\text{CH}_3}{\underset{\text{CH}_3}{\overset{\text{CH}_3}{\text{C}}}}^+$
- (B) $\text{CH}_3 - \text{CH}_2 - \overset{\oplus}{\text{CH}} - \text{CH}_2 - \text{CH}_3$
- (C) $\text{CH}_3 - \text{CH}_2 - \overset{\oplus}{\text{CH}_2}$
- (D) $\text{CH}_3 - \text{CH}_2 - \overset{\oplus}{\text{CH}} - \overset{\text{H}}{\underset{\text{CH}_3}{\text{C}}} - \text{CH}_3$

- Compound having general formula $\begin{array}{c} \text{R} \quad \text{OR} \\ \diagdown \quad \diagup \\ \text{C} \\ \diagup \quad \diagdown \\ \text{H} \quad \text{OR} \end{array}$ is called

- (A) diester (B) acid anhydride
 (C) hemiacetal (D) acetal

- The complex ion $[\text{Co}(\text{H}_2\text{O})_5(\text{ONO})]^{2+}$ and $[\text{Co}(\text{H}_2\text{O})_5(\text{NO}_2)]^{2+}$ are called:
 (A) linkage isomer (B) ionisation isomer
 (C) co-ordination isomer (D) geometrical isomer

- Inflammation of tongue is due to the deficiency of:

- (A) Vitamin B₁ (B) Vitamin B₂ (C) Vitamin B₃ (D) Vitamin B₆

- Identify the compound 'B' in the following series of reaction:



- (A) n-propyl chloride (B) Propanamine (C) n-propyl alcohol (D) Isopropyl alcohol

Hsc Chemistry Ii Board Paper 23

Siddappa N.Byrareddy

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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